



Punyashlok Ahilyadevi Holkar Solapur University

Criterion III – Research, Innovations and Extension

Key Indicator - 3.7 Collaboration

| Metric No. | |
|------------|--|
| 3.7.1 | <i>Number of collaborative activities with other institutions/ research establishment/industry for research and academic development of faculty and students during the year</i> |

3.7.1 Number of collaborative activities with other institutions/ research establishment/industry for research and academic development of faculty and students during the year

3.7.1.1: Total number of Collaborative activities with other institutions/ research establishment/industry for research and academic development of faculty and students during the year

| Sl. No. | Title of the collaborative activity | Name of the collaborating agency with contact details | Name of the participant | Year of collaboration | Duration | Nature of the activity | Link to the relevant document |
|---------|--|--|--|-----------------------|----------|------------------------|---|
| 1 | Research Activity | National Dong Hwa University, Taiwan | Dr.Sujit Kadam | 2021 | Continue | Research Paper | https://chemistry-europe.onlinelibrary.wiley.com/doi/abs/10.1002/slct.202002504 |
| 2 | Research Activity | Hanyang University, Korea | Dr.G.T.Chavan | 2022 | Continue | Research Paper | https://link.springer.com/article/10.1007/s11664-022-10019-9 |
| 3 | Research Activity | Mumbai University | Dr.R.C.Amhare | 2017 | Continue | Research Paper | https://chemistry-europe.onlinelibrary.wiley.com/doi/abs/10.1002/slct.202004750 |
| 4 | Research Activity | Pune University | Prof.H.M.Pathan | 2016 | Continue | Research Paper | https://www.sciencedirect.com/science/article/abs/pii/S1387700323006110X#::icj=The%20capacitive%20nature%20and%20internal%20state%20device%20AS%20SD |
| 5 | Research Activity | Pune University | Prof.P.D.More | 2020 | Continue | Research Paper | https://www.espublisher.com/uploads/article_pdf/esec8c785.pdf |
| 6 | Research Activity | Pune University | Prof. S.D.Sartale | 2020 | Continue | Research Paper | https://www.sciencedirect.com/science/article/abs/pii/S2352507X23000276 |
| 7 | Hydrous and amorphous cobalt phosphate thin-film electrodes synthesized by SILAR method for high-performing flexible hybrid energy storage devices | Department of Material science and Engineering | Mukund Mali | 2022 | 1 Year | Research Collaboration | https://pubs.acs.org/doi/abs/10.1021/acs.energyfuels.2c02202 |
| 8 | A Novel Recyclable Bi-Mg-O Composite Nano-Catalyst Promoted Rapid and Efficient Synthesis of Spiroindole and 4H-Pyran Derivatives | Department of Environmental Engineering, | Mukund Mali | 2022 | 1 Year | Research Collaboration | https://www.tandfonline.com/doi/abs/10.1080/10406638.2022.2124280 |
| 9 | Magnetically separable mixed metal oxide nanocomposite (Pd/MnFe2O4) for Suzuki cross-coupling in aqueous medium | Department of Chemical and Biological Engineering, Gachon University, 1342 Seongnam-aero, Sujung-gu, Seongnam-si, Gyeonggi-do 461-701, Republic of Korea | Mukund Mali | 2022 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/abs/pii/S0002328X22002893 |
| 10 | Interface engineering of polyoxotungstocobaltate with fern-leaf like BiVO ₄ microstructures for enhanced photocatalytic and supercapacitive performance | Department of Chemical and Biological Engineering, Gachon University, 1342 Seongnam-aero, Sujung-gu, Seongnam-si, Gyeonggi-do 461-701, Republic of Korea | Mukund Mali | 2022 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/abs/pii/S0927775723000584 |
| 11 | Multifunctional polyoxotungstocobaltate anchored fern-leaf like BiVO ₄ microstructures for enhanced photocatalytic and supercapacitive performance | Department of Chemical and Biological Engineering, Gachon University, 1342 Seongnam-aero, Sujung-gu, Seongnam-si, Gyeonggi-do 461-701, Republic of Korea | Mukund Mali | 2022 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/abs/pii/S0927775723000584 |
| 12 | Surface plasmon resonance based colorimetric probe for vitamin B1 detection: Applications to bio-fluid analysis | Fluorescence Spectroscopy Research Laboratory, Department of Chemistry, Shivaji University, Kolhapur, 416 004, Maharashtra, India | Mukund Mali | 2022 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/pii/S2666016423000853 |
| 13 | 3-Substituted-2-oxindole derivatives: Design, synthesis and their anti-tuberculosis and radical scavenging dual-action studies | The Material Science Research Institute, King Abdulaziz City for Science and Technology (KACST), Riyadh 11442, Saudi Arabia | Vikas Kadu, Raghunath B Bhosale | 2022 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/abs/pii/S0002328X22005737 |
| 14 | Green Synthesis, Molecular Docking, In Silico ADME and Biological Evaluation of Methoxy Substituted 1,5-Benzodiazepines as Potential Antioxidant, Anti-Inflammatory, and Antidiabetic Agents | Department of Pharmaceutical Chemistry, DKSS's Institute of Pharmaceutical Sciences and Research, Swami-Chincholi, Pune, Maharashtra, India | Raghunath B Bhosale | 2022 | 1 Year | Research Collaboration | https://www.tandfonline.com/doi/abs/10.1080/10406638.2021.1878244 |
| 15 | Syntheses, Molecular Docking and Biological Evaluation of 2-(2-hydrazinyl)thiazoles as Potential Antioxidant, Anti-Inflammatory and Significant Anticancer Agents | College of pharmacy vita Dist. Sangli | Raghunath B Bhosale, Anjana S Lawand, Vikas kadu | 2022 | 1 Year | Research Collaboration | https://www.ingentaconnect.com/content/ben/raiad/2022/000000016/00000002/4e100005 |
| 16 | Effect of embedding aluminium and yttrium on the magneto-optic properties of lanthanum spinel ferrite nanoparticles synthesised for photocatalytic degradation of methyl red | Department of Pharmaceutical Chemistry, College of Health Sciences, University of KwaZulu-Natal (Westville), Durban, 4000, South Africa | Sadanand N Shringare | 2022 | 1 Year | Research Collaboration | https://link.springer.com/article/10.1007/s10971-022-05951-5 |
| 17 | 3-Substituted-2-oxindole derivatives: Design, synthesis and their anti-tuberculosis and radical scavenging dual-action studies | Centre for Drug Discovery and Development, Col. Dr. Jeppiaar Research Park, Sathyabama Institute of Science and Technology, Rajiv Gandhi Road, Chennai, Tamil Nadu 600 119, India | Vikas Kadu, Raghunath B Bhosale | 2022 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/abs/pii/S0002328X22005737 |
| 18 | Synthesis, Molecular Docking and Biological Evaluation of 2-(2-hydrazinyl) thiazoles as Potential Antioxidant, Anti-Inflammatory and Anticancer Agent. | College of pharmacy vita Dist. Sangli | Raghunath B Bhosale, Anjana S Lawand, Vikas kadu | 2022 | 1 Year | Research Collaboration | https://www.ingentaconnect.com/content/ben/raiad/2022/000000016/00000002/4e100005 |
| 19 | Room-temperature ammonia gas sensor based on carboxylic acid-doped polyaniline | Department of Chemistry, Karmaveer Bhaurao Patil Mahavidyalaya (KBP), Pandharpur, 413304, Maharashtra, India | Anil A Ghanwat | 2022 | 1 Year | Research Collaboration | https://link.springer.com/article/10.1007/s00289-022-04215-0 |
| 20 | High-performance supercapacitive polyazomethines: Room temperature synthesis and their characterizations | Department of Physics, D.B.F. Dayanand College of Arts & Science, Solapur MS-413002, India | Anil A Ghanwat | 2022 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/abs/pii/S0002328X22002639 |
| 21 | Synthesis of Novel Hydrazones of Levofloxacin Related Molecule and their In Vitro Evaluation as Antioxidant, and Molecular Docking Studies | Wockhardt Ltd, Aurangabad | Anil A Ghanwat | 2022 | 1 Year | Research Collaboration | https://www.ingentaconnect.com/content/ben/raiad/2022/000000016/00000002/4e100008 |
| 22 | Development of new efficient and cost effective liquid-liquid extractive determination method for cobalt (II): Analysis of water, alloys and nano powder | Department of Chemistry, N. K. Orchid College of Engineering and Technology, Solapur, Maharashtra 413002, India | Anjana S Lawand | 2022 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/pii/S273050622000258 |
| 23 | In Vitro Anticancer Screening, Molecular Docking and Antimicrobial Studies of Triazole-Based Nickel(II) Metal Complexes | Chemistry Research Laboratory, Department of Chemistry, Shri Shivaji Mahavidyalaya, Solapur 413 411, Maharashtra, India | Umesh Barache | 2022 | 1 Year | Research Collaboration | https://www.mdpi.com/1420-3049/27/19/6548 |
| 24 | Analytical optimization of liquid-liquid extractive spectrophotometric assessment protocol for tetravalent platinum: Analysis of environmental samples and cisplatin | Materials Research Laboratory, Department of Chemistry, C.B. Khedgi's Basaveshwar Science, Raja Vijaysinh Commerce and Raja Jaysinh Arts College, Akkalakot 413216, Maharashtra, India | Anjana S Lawand | 2022 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/abs/pii/S1386142522010666 |
| 25 | Potential arsenic-chromium-lead Co-contamination in the hilly terrain of Arunachal Pradesh, north-eastern India: Genesis and health perspective | Escuela de Ingeniería y Ciencias, Tecnológico de Monterrey, Campus Monterrey, Monterrey 64849, Nuevo Leon, Mexico | Umesh Barache | 2023 | 1 Year | Research Collaboration | https://www.sciencedirect.com/science/article/abs/pii/S004956532300933X |
| 26 | Cyber Security Awareness: A Movement of Digital Literacy Towards making of Digital India | Bharathi Vidyaapeeth, Solapur | Dr. Shiram D. Raut | 2022 | 1 | Research Paper | https://www.indianjournalofmanagement.com/index.php/ibmd/article/view/172617 |
| 27 | Cyber Security Awareness: A Movement of Digital Literacy Towards making of Digital India | Bharathi Vidyaapeeth, Solapur | Dr. Ashok R. Shinde | 2022 | 1 | Research Paper | https://www.indianjournalofmanagement.com/index.php/ibmd/article/view/172617 |
| 28 | Acupressure Consultancy Program | School of Allied Health Sciences and Shree Sai Samarth Educational and Research Society Nagpur | 39 | Dec. 2020 | 3 Years | Consultancy | NIL |
| 29 | Summer Internship Program | Maharashtra University of Health Sciences, Nashik | 10 | Jan. 2018 | 5 Years | Internship | NIL |

Spray Synthesized Mn-doped CuO Electrodes for High Performance Supercapacitor

Pravin More,^[a] Sujit A. Kadam,^[b] Yuan-Ron Ma,^[b] Yan-Ruei Chen,^[c] Nilesh Tarwal,^[d] Yuvraj Navale,^[e] Amol Salunkhe,^[e] and Vikas Patil^[e]

A pristine copper oxide and manganese-doped copper oxide nanostructured films were prepared by simple spray pyrolysis technique. A study of the effects of Mn doping with three different concentrations as 2, 4, and 6 wt% were studied in the view of enhancement of supercapacitive performance of copper oxide electrode. The 6 wt% manganese-doped copper oxide thin film changed their surface morphology considerably

favorable to the exchange of ions in the Na_2SO_4 electrolyte. The electrochemical properties of the pristine copper oxide and manganese-doped copper oxide were investigated in 1 M Na_2SO_4 electrolyte within the potential window of -1.0 to 0.4 V. A maximum specific capacitance of 801.61 F/g was obtained for the manganese-doped copper oxide thin film electrode.

Introduction

Supercapacitors (SCs), also known as electrochemical capacitors, are one of the most promising candidates. It exhibits attractive characteristics such as high power density, permanent charge-discharge cycle, low series resistance, longer life cycle, and wide range of operating temperature, easily maintenance, and environmental friendliness than secondary batteries.^[1–4] The charges are stored in supercapacitors via two principles: diffusion controlled kinetics (pseudocapacitors) and surface-controlled kinetics (electrical double-layer supercapacitor).^[5,6] Currently transition metal oxides electrodes have a great choice as sophisticated supercapacitor due to transition metal oxides abundant availability, environmentally friendly, large surface area as well as high theoretical specific capacitance.^[7,8] Although, transition metal oxides/hydroxides for example, RuO_2 , V_2O_5 , NiO , $\text{Ni}(\text{OH})_2$, Co_3O_4 , $\text{Co}(\text{OH})_2$, MnO_2 were used to achieve excellent supercapacitive performance.^[8,9] However RuO_2 is restricted for its commercial use due to expensive, toxic, and less abundant.^[8] Among aforesaid, it is possible for transition metal oxides to have a variety of

oxidation states which can enhance redox charge transfer, resulting in much higher specific capacitances.^[10] A promising electrode material, manganese oxide has a high specific capacitance, an ideal charge-discharge curve, is readily available, and is stable. CuO is an attractive choice for energy storage since it is low cost, low toxicity, abundant, environmentally friendly, and has a large surface area and high conductivity.^[10] Manganese oxides have been found as one of the most promising candidate for supercapacitor applications due to their natural abundance, low cost.^[11] Based on above considerations, we designed a pristine and Mn-doped CuO electrodes by spray pyrolysis technique. Pang et al. synthesized MnO_2 and reported maximum specific capacitance of 689 F/g at a current density of 10 mA/cm^2 in 6 M KOH electrolyte.^[11] More et al. deposited Mn_3O_4 and found maximum specific capacitance of 527 F/g at a scan rate of 5 mV/s in $1 \text{ M Na}_2\text{SO}_4$ electrolyte.^[11] Xia et al. synthesized MnO_2 and reported maximum specific capacitance of 364 F/g at 10 mV/s in $1 \text{ M Na}_2\text{SO}_4$ electrolyte.^[12] In the research community, researchers have focused on a synthesis of mixed metal oxides than single metal oxide to achieve the desired performance of supercapacitor. In addition to that following unique features of metal oxides can breakthroughs for supercapacitors such as i. co-existence of two different metal oxides in a single crystal structure could improve the electrical conductivity than single one, ii. nanoporous metal oxides can provide a higher surface area.^[7] Durai et al. synthesized a pristine and Mn-doped CuO electrodes by a reactive radio frequency magnetron sputtering technique and reported maximum specific capacitance of 10 and 81.80 mF/cm^2 at a scan rate of 10 mV/s in 3 M KOH electrolyte, respectively.^[2] Rakic et al. prepared CuO and MnO_2 - CuO nanocomposite electrodes and showed maximum specific capacitance of 191.06 and 279.12 F/g , respectively.^[10] Suresh et al. synthesized CuO and $5 \text{ wt\% Mn-doped CuO}$ and reported maximum specific capacitance of 375 and 625 F/g in 2 M KOH electrolyte at 2 mV/s , respectively.^[13] Shinde et al. deposited $\text{CuO/Cu}(\text{OH})_2$ and Mn: $\text{CuO/Cu}(\text{OH})_2$ hybrid electrodes at various Mn concentrations. As a result $\text{CuO/Cu}(\text{OH})_2$ and 3%

[a] Dr. P. More

Department of Physics, Ahmednagar College, Ahmednagar, 414001, M.S., India

E-mail: pravin.more@aca.edu.in

[b] S. A. Kadam, Prof. Y.-R. Ma

Department of Physics, National Dong Hwa University, Hualien, 97401, Taiwan

E-mail: ksujit17@gmail.com

[c] Y.-R. Chen

Department of Engineering and System Science, National Tsing Hua University, Hsinchu, Taiwan

[d] Dr. N. Tarwal

Smart Materials Research Laboratory, Department of Physics, Shivaji University, Kolhapur, 416004, M.S., India

[e] Dr. Y. Navale, A. Salunkhe, Prof. V. Patil

Functional Materials Research Laboratory, School of Physical Sciences, Punyashlok Ahilyadevi Holkar Solapur University, Solapur, 413255, M.S., India

Supporting information for this article is available on the WWW under <https://doi.org/10.1002/slct.202202504>





Mn-Incorporated α -Fe₂O₃ Nanostructured Thin Films: Facile Synthesis and Application as a High-Performance Supercapacitor

Sujit A. Kadam¹ · Yuan-Ron Ma¹ · Yan-Ruei Chen² · Yuvraj H. Navale³ · Amol S. Salunkhe³ · Vikas B. Patil³ · Sachin D. Ralegankar⁴ · Pravin D. More⁴

Received: 9 June 2022 / Accepted: 3 October 2022
© The Minerals, Metals & Materials Society 2022

Abstract

Among all the transition metal oxides, iron oxide-based materials are excellent for supercapacitor performance. Here, Mn-incorporated α -Fe₂O₃ (Mn: α -Fe₂O₃) nanostructured thin films (with 3%, 5%, and 7% Mn) are prepared via spray pyrolysis. All the synthesized nanostructured thin films are characterized by x-ray diffraction (XRD), optical study, Fourier transform infrared spectroscopy (FTIR), field emission scanning electron microscopy (FESEM), transmission electron microscopy (TEM), and contact angle for the structural, optical, morphological and wettability analysis, respectively. The band gap of Mn: α -Fe₂O₃ nanostructured thin films is tuned by changing Mn concentration. The increasing Mn concentration shifts the valance band edge towards the conduction band edge, reducing the band gap. The linear band gap decrease of 0.44 eV with the addition of Mn concentration, along with the band gap reduction, affects supercapacitive performance. The prepared 7% Mn: α -Fe₂O₃ nanostructured electrode exhibits excellent specific capacitance of 688.6 F g⁻¹ at a scan rate of 5 mV s⁻¹ in 1 M Na₂SO₄ electrolyte, energy density (6 Wh kg⁻¹), and power density (12 kW kg⁻¹) at a current density of 5 mA g⁻¹.

Keywords Spray pyrolysis technique · Mn: α -Fe₂O₃ nanostructure · contact angle · electrochemical analysis · supercapacitor

Introduction

In recent years, a multiplicity of energy storage devices, viz. capacitors, supercapacitors, and batteries, are accessible in daily life. However, the consequences of such energy storage devices primarily depends not only on the effectiveness but also on the stability of the electrode resources.¹ In pursuit of more advanced storage devices, highly requested research efforts are being made in modern society for effecting large-scale employability in the area of durable energy storage devices. The currently available supercapacitors are the best

devices since they act as a bridge between the traditional capacitor and secondary batteries. The supercapacitors are characterized by a high energy density, a long life cycle, excellent rate capabilities, a wide operating temperature range, enhanced safety, efficiency, and good endurance.^{2,3} The charge storage mechanisms in supercapacitors are based on two principles: one is the pseudocapacitor mechanism which relies on reversible redox reaction. The other is the electrostatic adsorption ions at the electrode/electrolyte interface and it possesses low energy density.^{1–4} A supercapacitor stores charge at the electrodes; it can be charged and discharged at a higher rate and can undergo longer cycles than a battery.⁵ Supercapacitors have the potential to be used in portable electronic devices and power hybrid cars.

A transition metal oxide-based supercapacitor electrode exhibits a specific capacitance that is 10–100 times higher than carbon-based materials.⁶ Electrode materials with a large surface area have a high specific capacitance, while nanoscale materials tend to have higher surface area which increases electrode–electrolyte contact and enhances charge transfer reactions. A metal oxide such as iron oxide has been widely used in pigments, catalysts, sensors, environmental pollutant agents, biomedical materials and electrode

✉ Pravin D. More
drpravinmore@gmail.com

¹ Department of Physics, National Dong Hwa University, Hualien 97401, Taiwan
² Department of Engineering and System Science, National Tsing Hua University, Hsinchu, Taiwan
³ Functional Materials Research Laboratory, School of Physical Sciences, Punyashlok Ahilyadevi Holkar Solapur University, Solapur 413255, MS, India
⁴ Department of Physics, Ahmednagar College, Ahmednagar 414001, MS, India



Influence of the Camphor Sulphonic Acid (CSA) Intercalation on the Micro-structural and Gas Sensing Properties of Polyaniline-CeO₂ Nanohybrid for NH₃ Gas Detection

Reshma N. Dhanawade,^[a] Nanasaheb S. Pawar,^[a] Manik A. Chougule,^[a] Gajanan M. Hingangavkar,^[a] Tanaji M. Nimbalkar,^[a] Ganesh T. Chavan,^[b] Chan-Wook Jeon,^[b] and Vikas B. Patil^{*[a]}

Developing a high performance sensing materials operating at room temperature (30 °C) is eminently a challenging task. The facile chemical oxidative polymerization route was employed for the synthesis of PANi-CeO₂ nanohybrids with CSA intercalation (10–50 wt%) on the chains of protonated PANi. The cubic crystal structure of PANi-CeO₂-CSA nanohybrids were revealed by X-ray diffractometry (XRD). Scanning electron microscopy (SEM) and transmission electron microscopy (TEM) were employed for the structural and morphological investigations. X-ray Photoelectron Spectroscopy (XPS) and Raman spectro-

scopy confirmed the formation of PANi-CeO₂-CSA nanohybrid. The consequences of CSA intercalation on the gas sensing performance of PANi-CeO₂ nanohybrids are explored through custom designed sensing system. The gas sensing investigations revealed that PANi-CeO₂-CSA (50 wt%) nanohybrid exhibited highest response (93%) towards 100 ppm NH₃ at 30 °C. Advantageously, CSA intercalated PANi-CeO₂ nanohybrid gas sensor delivered fast response and quick recovery (8 sec and 482 sec) with admirable stability (84.09%) towards NH₃ at 30 °C.

Introduction

Reproducible, inexpensive and highly active gas sensors are considered significant for versatile applications such as manufacturing, environmental, health and military operations to avoid impending hazards for human being and environment.^[1–3] Ammonia (NH₃) is a toxic and combustible contaminant, posturing continuous risks to human health and ecology if it reaches to 25 ppm in 8 hrs.^[4] NH₃ spoils the ecosystem because of natural and artificial sources like industrialization,^[5] automobile exhaust^[6] and household activities. Therefore, an excessive efforts have been carried out for developing NH₃ gas sensor that is miniaturized, convenient and highly sensitive towards harmful NH₃ gas at room temperature for human being and environmental protection. Many gas sensing materials are available, such as conducting polymers,^[7,8] semiconducting metal oxides (SMO)^[9,10] and carbon-based materials,^[11,12] differ in their behaviors and gas sensing performances. Nowadays, semiconducting metal oxides such as cerium oxide (CeO₂), tungsten oxide (WO₃), iron oxide (Fe₂O₃) and tin oxide (SnO₂) are used in gas sensors because of

excellent stability and improved gas sensing properties. Cerium oxide (CeO₂) is n-type semiconductor metal oxide with outstanding electrical conductivity and excellent redox reactions that improve gas sensing characteristics.^[13,14] On the other hand, cerium oxide (CeO₂) comprising high operating temperature with heat that induced crystal growth and hazard of explosion, it restricts the practical applications.^[15] Conducting polymers are more reliable candidates due to their low price, most significant selectivity, fast response and recovery time at room temperature as compared with metal oxides.^[16] Amongst the conducting polymers, PANi is more efficient for revelation of toxic NH₃ gas at 30 °C.^[17,18] Moreover, pristine PANi is simple to synthesize and it comprises with distinct molecular structure and excellent stability. But, it has some drawbacks, including low selectivity, long-term stability and extended response and recovery time.^[19] The gas sensing performances of cerium oxide and pristine PANi are inconvenient and they constrained by a number of factors. Therefore, for the achievement of excellent gas sensing performance, the composite material comprising with organic and inorganic material properties such as PANi-CeO₂ nanohybrid was synthesized.

In the present research work, CSA intercalated PANi-CeO₂ nanohybrid was synthesized utilizing a facile polymerization process to improve the gas sensitivity, selectivity, stability and response-recovery time.^[20,21] The role of CSA is a potential dopant, which increases the conductivity as well as stability of PANi-CeO₂ nanohybrids.^[22,23] Therefore, CSA intercalated PANi-CeO₂ nanohybrids delivers excellent gas response towards NH₃ at 30 °C. The related investigations for gas sensors were studied. Navale et.al^[22] described the fabrication of CSA (10–50 wt%) intercalated PPy-Fe₂O₃ hybrid nanocomposites for



[a] R. N. Dhanawade, N. S. Pawar, Dr. M. A. Chougule, G. M. Hingangavkar, T. M. Nimbalkar, Prof. Dr. V. B. Patil
Functional Materials Research Laboratory, School of Physical Sciences
Punyashlok Ahilyadevi Holkar Solapur University
413 255 Solapur, Maharashtra, India
E-mail: vbpatil@sus.ac.in

[b] Dr. G. T. Chavan, Prof. C.-W. Jeon
School of Chemical Engineering,
Yeungnam University,
38541 Gyeongsan, Gyeongbuk, Republic of Korea



Short communication

Nanoarchitectonics of Bi_2CuO_4 electrodes for asymmetric $\text{Bi}_2\text{CuO}_4//\text{AC}$ solid-state device in supercapacitor application

R.G. Bobade^a, U.T. Nakate^a, P. Raasiah^b, M. Ouladsmoute^a, B.J. Lokhande^a, R.C. Ambare^a  Show more  Share  Cite<https://doi.org/10.1016/j.inoche.2021.110998>

Get rights and content

Abstract

The fashionable investigates the bismuth oxide nano-materials were synthesized using bismuth nitrate as an ingredient. The samples were deposited using the electrodeposition technique via an aqueous route with varying deposition time variations. Prepared samples were annealed at 573K in a muffle furnace. All deposited thin film electrodes were physical and electrochemical characterized via X-ray diffraction (XRD), Field emission scanning electron microscopy (FE-SEM), Elemental mapping, Energy dispersive X-ray spectroscopy (EDX), Contact Angle, X-ray photoelectron spectroscopy (XPS) and supercapacitive measurements. XRD of all deposited samples reveals polycrystalline nature with a tetragonal crystal structure. FE-SEM shows revolutionary morphological images of spruce leaf-like architecture. EDX analysis shows the elemental mapping of the deposited materials. TEM reveals the formation of nano granules structures indicating the formation of the crystalline bismuth oxide. XPS confirms the formation of Bi_2CuO_4 . All samples show the hydrophilic nature of deposited materials from contact angle. Cyclic voltammetry (CV), Galvanostatic charge-discharge (GCD), and impedance measurements are used to examine the supercapacitive characteristics in an aqueous 1M KOH electrolyte. All CV curves show mixed capacitive nature. The results of the investigations are highest specific capacitance (SC) of 706.47F/g at a scan rate of 2mV/s was calculated. The specific energy (SE) of 49.89 Wh/kg, and specific power (SP) of 17.51 kW/kg were calculated at 40mA/cm² current density. The capacitive nature and internal resistance were observed using the Nyquist plot for the optimized electrode which is about 2.11 Ohm. CV, GCD, and impedance measurements are used to characterize the electrochemical parameters of the asymmetric supercapacitor.





Preparation of Magnesium Oxide (MgO) Thin Films by Spray Pyrolysis and Its Capacitive Characterizations

M. T. Mhetre, H. M. Pathan, A. V. Thakur and B. J. Lokhande*

Abstract

Magnesium oxide (MgO) electrodes for the supercapacitor applications have been prepared using automatic spray pyrolysis technique. 100 ml of 1 M aqueous $\text{Mg}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ solution was sprayed at 2 ml/min. on to the stainless steel (SS) substrate heated to 673 K. The prepared electrodes were analyzed structurally by X-ray Diffraction (XRD) technique, field emission scanning electron microscope (FESEM) analysis, energy dispersive X-ray analysis (EDX), and contact angle measurement, functional group by Fourier transform infrared spectroscopy (FTIR) analysis and electrochemically by cyclic voltammetric analyses (CV), electrochemical impedance spectroscopy (EIS) and the galvanostatic charge discharge (GCD) technique. Different peaks in XRD pattern coincide with the standard database of JCPDS 01-1235 indicating the formation of MgO. The formation of crystalline MgO has further been corroborated by EDX. The formation of MgO was studied the molecular interactions between the metal, oxygen and carbon elements through FTIR analysis. Peaks in the CV curves confirm the redox behavior of the MgO during the CV. The measured specific capacitance (SC) was as high as 202.36 Fg^{-1} . Superb electrochemical stability is observed at 100 mV/s scan rate. Using GCD analyses show different behavior from the normal capacitors. The charging and discharging times were found nearly same for different applied currents. The observed maximum specific capacitance (SC) was 163.55 Fg^{-1} which is comparable with that obtained by CV.

Keywords: Magnesium Oxide; Spray pyrolysis; Contact angle; Supercapacitive study.
Received: 26 April 2022; Revised: 19 September 2022; Accepted: 15 October 2022.
Article type: Communication.

1. Introduction

Climate change and fossil fuel depletion have triggered intense scientific research to explore renewable and sustainable energy sources; demands from hybrid electric vehicles and pulsed power systems also promote the development of alternative energy.^[1–4] The supercapacitor is one of the promising devices to meet the ever-growing need, with its higher energy density than that of common capacitors and better power density than that of batteries.^[5] Exploring electrode materials with high specific capacitance (SC) is necessary to improve the energy density of supercapacitors. Carbon materials, conducting polymers and transition metal oxides/hydroxides with high surface area are commonly used as supercapacitor electrode materials.^[6] Carbon based materials (carbon aerogels, carbon nanotubes and graphene)^[7–9] as electrode materials for electrical double-layer capacitors have been extensively studied to increase their low SC which directly limited their practical application.^[10] Conducting

polymers such as polyaniline, polypyrrole, poly(3,4-ethylene dioxy thiophene), etc. have significant drawbacks of low rate of charge-discharge due to slow ion diffusion within the bulk of the electrode.^[11] Transition metal oxides/hydroxides^[12] /sulphides^[13] are promising candidates owing to their low environmental toxicity and high SC, which have been widely investigated in an attempt to attain high SC and long cycle life.

Pseudocapacitors/redox capacitors store the charge via fast faradaic reactions. Redox capacitors are used for high power applications due to quick discharging.^[14–16] Variety of materials including not only electrically conducting polymers in pure^[17,18] and composite/hybrid forms^[19–22] but plenty of transition-metal oxides (TMO), RuO_2 ,^[23] Co_3O_4 ,^[24] Fe_2O_3 ,^[25] NiO ,^[26] Fe_3O_4 ,^[27] FeOOH ,^[28,29] Cu(OH)_2 , NiCo_2O_4 , MnCo_2O_4 ^[30–32] etc., are being tested for fabrication of redox capacitors. Ubiquitous need of alternatives for RuO_2 resulted due to high cost and limited sources of Ru. Although, due to large bandgap of MgO (~7.8 eV) the electrochemical charge storage capability of the MgO is quite less. However, even not belonging to the transition metal oxide group, the MgO has been reported for improving the stability of the electrochemical energy storage electrode materials studied by K. Karthikeyan *et al.*^[33] Also, it produces

Lab of electrochemical studies, School of Physical Sciences, P.A.H. Solapur University, Solapur

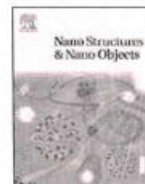
*E-mail: bjl@khande@yahoo.com (B. J. Lokhande)





Contents lists available at ScienceDirect

Nano-Structures & Nano-Objects

journal homepage: www.elsevier.com/locate/nanosoIntercalation of two-dimensional graphene oxide in WO₃ nanoflowers for NO₂ sensingGajanan M. Hingangavkar^{a,b}, Sujit A. Kadam^c, Yuan-Ron Ma^c, Shrikrishna D. Sartale^d, Ramesh N. Mulik^a, Vikas B. Patil^{b,*}^a Department of Physics, DBF Dayanand College of Arts & Science, Solapur, (MS), 413002, India^b Functional Materials Research Laboratory, School of Physical Sciences, PAH Solapur University, Solapur, (MS), 413255, India^c Department of Physics, National Dong Hwa University, Hualien 97401, Taiwan^d Department of Physics, Savitribai Phule Pune University, Pune, (MS), 411007, India

ARTICLE INFO

Article history:

Received 14 November 2022

Received in revised form 20 January 2023

Accepted 1 March 2023

Keywords:

WO₃-GO nanohybrid

Hydrothermal

XPS

BET

NO₂ sensor

ABSTRACT

Chemiresistive WO₃-GO (WG) nanohybrid sensors were designed by a cost-effective hydrothermal process. The structure, morphology, composition and surface features of the prepared WG nanohybrids were explored by an array of analytical techniques, such as X-ray diffractometry (XRD), Raman spectroscopy, X-ray photoelectron spectroscopy (XPS), field emission scanning electron microscopy (FESEM), energy dispersive X-ray analysis (EDAX), high-resolution transmission electron microscopy (HRTEM) and Brunauer–Emmett–Teller (BET) surface profiling. The gas sensing performance of the WG nanohybrids reveals that the small amount of GO significantly impacts the sensor performance. The enhanced gas sensing performance of the WG nanohybrid sensor with a GO content of 3 weight% exhibits an excellent response to 100 ppm of NO₂, attaining 239%, which is nearly fourfold higher than that of pristine WO₃ (61%) at 150 °C, and shows outstanding selectivity, reproducibility and stability (84.5%). Impedance spectroscopy was employed to understand the interaction between the NO₂ gas molecules and the WG nanohybrid.

© 2023 Elsevier B.V. All rights reserved.

1. Introduction

A rapid economic boom throughout the early decades fueled the speedy increase in global energy consumption. Energy is mainly obtained from the burning of fossil fuels. As a result, the environment gets contaminated with poisonous gases released from burning fossil fuels. Pollutant gases emitted by the combustion of fuels, such as nitrogen dioxide (NO₂), carbon monoxide (CO), hydrogen disulfide (H₂S) and others, pollute the atmosphere. These are harmful to both living organisms and humans. In a short period, exposure to NO₂ (≥ 10 ppm) may lead to fast nasal and throat irritation, as well as edema and distress. NO₂ concentrations exceeding 100 ppm cause death by suffocation when inhaled within a short time frame. Additionally, NO₂ causes photochemical smog, acid rain and accelerates microscopic particle formation. Thus, for environmental and health monitoring, cost-effective and simple chemiresistive sensors with outstanding NO₂ sensing capability are of the utmost importance [1–4].

Transition metal oxides, such as WO₃, TiO₂, SnO₂, ZnO, and Fe₂O₃ [5,6], have been extensively studied in the last few decades

for NO₂ detection because of their semiconducting nature. Y.H. Navale et al. have reported a CuO nanocube sensor for NO₂ detection using the thermal evaporation technique [2]. Lizhai Zhang et al. have worked on a CeO₂/Graphene heterostructure for NO₂ sensing using a hydrothermal approach [1]. Hongwen Zhang et al. have explored WO₃ decorated using Au nanoparticles for the selective detection of NO₂ gas [4]. Shubin Sun et al. have synthesized nanocomposites of graphene nanoplatelets and WO₃·H₂O using a simple precipitation method [7]. WO₃ nanoflowers (NFs) are 3D nanostructures [8–17] with an ability to detect NO₂ gas that have been extensively investigated over a few decades, attributable to their low cost, excellent chemical stability, crystal structures and superior sensitivity. It is recognized that the effect of gas-sensing of WO₃ NFs is initiated after the interaction among oxygen ionic species (O²⁻, O₂⁻ and O⁻) adsorbed on the nanostructured surfaces and the gas molecules [7,18].

The inherent properties of WO₃ NFs, such as absorption of low visible light and fast recombination of carriers, reduce its utility in some applications, such as photocatalysis and gas sensing. We can straightforwardly overcome the shortcomings of pristine WO₃ NFs by hybridization with nanomaterials of the carbon family (carbon nanotubes and graphene) to improve their performance [19,20]. Graphene oxide (GO) is the most impressive graphene-based structure, because of its exclusive surface

* Corresponding author.

E-mail address: vbpatil@sus.ac.in (V.B. Patil).

ADVERTISEMENT

[RETURN TO ISSUE](#)[< PREV](#) **BATTERIES AND ENERGY...** [NEXT >](#)[Get e-Alerts](#)

Hydrous and Amorphous Cobalt Phosphate Thin-Film Electrodes Synthesized by the SILAR Method for High-Performing Flexible Hybrid Energy Storage Devices

Vinod V. Patil, Sachin S. Pujari, Shraddha B. Bhosale, Sambhaji S. Kumbhar, Vinayak G. Parale, Jayavant L. Gunjikar, Jyoti Ho. Park, Chandrakant D. Lokhande, Mukund G. Mali, Dattakumar S.

Sachin S. Pujari

Centre for Interdisciplinary Research (CIR), D. Y. Patil Education Society (Deemed to be University), Kolhapur 416006, India

More by Sachin S. Pujari

✓ **Cite this:** *Energ*

Publication Date: (

<https://doi.org/10.>

Copyright © 2022

[Request reuse perm](#)

Article Views

432

Altmetric

-

Citations

1

[LEARN ABOUT THESE METRICS](#)

Share Add to Export



Access Through
Indian Institute of Technology P...

 [Add or Change Institution](#)

Other access options

[Pair your account to your Institution](#) 

 [Supporting Info \(1\) »](#)

SUBJECTS: [Capacitors](#), [Cobalt](#), [Electrical properties](#), [Electrodes](#), [Phosphates](#)



Log in

| Register



Cart

Home ▶ All Journals ▶ Polycyclic Aromatic Compounds ▶ List of Issues ▶ Volume 43, Issue 7
▶ A Novel Recyclable Bi-Mg-O Composite Nan ...

Polycyclic Aromatic Compounds >

Volume 43, 2023 - Issue 7

207 | 0

Views | CrossRef citations to date | Altmetric

Research Articles

A Novel Recyclable Bi-Mg-O Composite Nano-Catalyst Promoted Rapid and Efficient Synthesis of Spirooxindole and 4H-Pyran Derivatives

Abhijeet Mulik, Dae Sung Lee, ...

Pages 6665-6678 | Received: 25 Sep 2022

Vishvanath Ghanwat

b Department of Chemistry, Yashwantrao Chavan Institute of Science, Satara, India

Mukund Mali, D.-Y. Kim,

Received: 25 Sep 2022

“ Cite this article

View further author information

Check for updates

Sample our
Physical Sciences
Journals

>> Sign in here to start your access
to the latest two volumes for 14 days



Full Article

Figures & data

References

Supplemental

Citations

Metrics

Reprints & Permissions

Read this article

Abstract





A novel Bi-Mg-O composite nano-catalyst was prepared by using $\text{Bi}(\text{NO}_3)_3$ and nano-MgO. Prepared catalyst was characterized using x-ray diffraction (XRD) analysis, scanning electron microscope (SEM), X-ray photoelectron spectroscopy (XPS) technique which confirms its nano-crystalline nature and composition. The synthesized novel catalyst Bi-Mg-O was found to be competent catalyst for the synthesis of spirooxindole derivatives by multicomponent reaction of isatin, dimedone, malononitrile and 4H-pyran derivatives by reaction of dimedone, aryl



Colloids and Surfaces A: Physicochemical and Engineering Aspects

Volume 662, 5 April 2023, 130974

Multifunctional polyoxotungstocobaltate anchored fern-leaf like BiVO₄ microstructures for enhanced photocatalytic and supercapacitive performance

Gopal Mali ^a, Laxman Walekar ^b, Nagesh Kolhe ^b, Abhijit N. Kadam ^{c, f}, Rohan Kore ^d, Dattakumar Mhamane ^e  ,
Harichandra Parbat ^f, Sang-Wha Lee ^c, Balkrishna Lokhande ^d, Vaishali Patil ^g, Gavisiddapa Gokavi ^a, Mukund Mali ^b  

[Show more](#) 

 Share  Cite

<https://doi.org/10.1016/j.colsurfa.2023.130974> 

[Get rights and content](#) 

Abstract

In this study, a simple sonochemical approach for molecular anchoring of CoW₁₂O₄₀ [12-tungstocobaltate (II)] on fern-leaf like BiVO₄ microstructures was employed. The various compositions were prepared by varying contents (0.5%, 1%, 1.5%, 2% and 5%) of CoW₁₂O₄₀ (hereafter denoted as CoWO). The as-synthesized samples were characterized and confirmed by various physicochemical tools. Furthermore, the performance of resulting composites was tested toward visible light driven photocatalytic treatment of organic pollutants and supercapacitor. Among them, 1CoWO/BVO composites showed optimal performance as compared to other composites and bare fern-leaf like BiVO₄. To be specific, the photocatalytic performance of 1CoWO/BVO showed 97% and 78% degradation of methylene blue dye (MB) and tetracycline drug (TC), respectively. Additionally, the reusability of optimal 1CoWO/BVO photocatalyst was confirmed by performing its five consecutive runs towards degradation of MB and TC. Moreover, 1CoWO/BVO electrode yielded the specific capacitance of 245 Fg⁻¹ at 2 mV/s scan rate, supporting its applicability as an efficient supercapacitor electrode. The synergistic effect of CoWO and BiVO₄ dominated their bare counterparts towards photocatalytic as well as electrochemical performance. Thus, this work rationally opens up the avenues to employ POMs based novel hybrid materials for future environmental and energy storage application.

Graphical Abstract







Case Studies in Chemical and Environmental Engineering

Volume 8, December 2023, 100380

Case Report

Surface plasmon resonance based colorimetric probe for vitamin B₁ detection: Applications to bio-fluid analysis

Uttam R. Kondekar ^a, Laxman S. Walekar ^b, Samadhan P. Pawar ^{a, c}  , Mukund G. Mali ^b, Dilip D. Anuse ^d, Chandrakant G. Gardi ^e, Shamkumar Deshmukh ^f, Prashant V. Anbhule ^a, Govind B. Kolekar ^a  

[Show more](#) 

 Outline |  Share  Cite

<https://doi.org/10.1016/j.cscee.2023.100380> 

[Get rights and content](#) 

Under a Creative Commons [license](#) 

[open access](#)

Abstract

This study reports simple analytical approach for thiamine (Vitamin B₁) detection based on induced aggregation and alternation in colorimetric properties of gold nanoparticles (AuNPs), which was synthesized through citrate reduction approach. Furthermore, the citrate capped AuNPs are characterized by various analysing tools. The addition of thiamine persuades the aggregation of citrate-AuNPs and further leading to red to blue colour transition with decrease in absorbance intensity. The proposed method achieves good linearity with a correlation coefficient of 0.9843. By using our proposed strategy, thiamine was detected by unassisted vision as well as absorption spectroscopy. Under the most favorable condition method achieves good linear relationship between concentration range 0.01–0.8 $\mu\text{g mL}^{-1}$ with limit of detection of 0.0067 $\mu\text{g mL}^{-1}$. Under the premium condition, the method offers excellent selectivity towards thiamine detection in presence of different interfering species. Further practical applicability of the method was checked by using blood serum and urine sample via standard addition method. The obtained recoveries were acceptable in the range of 98.70–102.97% for added thiamine concentration. Thus, the proposed method may emerge as a target specific and highly sensitive tool towards thiamine detection.

 Previous

Next 

Keywords

Gold nanoparticles; Thiamine; Aggregation; Colour change; Bio-fluids; SPR; Absorption

1. Introduction

Vitamins are one of the most required micronutrient for human body to work properly. Anabolic processes do not produce some vitamins, food and medicines are good candidate to provide it [1]. B complex is a combination of Vitamin B₁, B₆ and B₁₂ and important participant in various functions of metabolism and excess or deficiency of



3-Substituted-2-oxindole derivatives: Design, synthesis and their anti-tuberculosis and radical scavenging dual-action studies

Mahesh Hublikar^a, Vikas Kadu^a, Dattatraya Raut^a, Sachin Shirame^a, Sivaraj Anbarasu^b, Muhanna K. Al-Muhanna^c, Parameshwar Makam^{d,e}  , Raghunath Bhosale^a 

[Show more](#) 

 Share  Cite

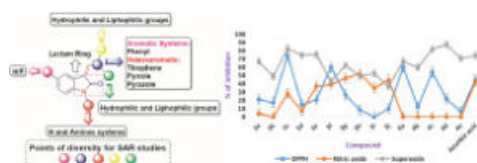
<https://doi.org/10.1016/j.molstruc.2022.132903> 

[Get rights and content](#) 

Abstract

Despite the fact that the oxindole has long been thought to be the best structural scaffold, research shows that oxindole derivatives are being studied the least against tuberculosis (TB), a historically airborne illness. The present research focuses on the development of 3-substituted-2-oxindole derivatives with a combination of structural and functional alterations in order to create dual-active new chemical entities with anti-TB, anti-oxidant, and radical-scavenging properties. The physicochemical properties of these compounds are in line with the theories of Lipinski, Veeber, and Leeson. SWISSADME was used to assess the drug-like molecular nature and pharmacokinetics features. Multistep synthetic procedures were used to synthesise the chemicals. Based on *in vitro* anti-TB and radical scavenging property analyses, the synthesised 3-substituted-2-oxindole derivatives were confirmed as dual-active compounds. The synthesised 3-substituted-2-oxindole derivatives were confirmed to possess the dual therapeutics action. The discovery of radical scavengers **4d**, **4a**, **4c**, **3d**, **3c**, **3f**, **3e**, **3h**, **3g**, and **3j** that are more potent than Ascorbic acid, the standard. The molecules, **3c**, **3j**, **3b**, and **4b** have exhibited moderate to active anti-TB properties with a % reduction values of 46.42%, 40.89%, 39.76% and 39.32%, respectively. The findings suggest that fine-tuning molecules might result in compounds with better anti-TB and radical-scavenging properties.

Graphical abstract



[Download : Download high-res image \(117KB\)](#)

[Download : Download full-size image](#)



Log in

| Register



Cart

Home ▶ All Journals ▶ Polycyclic Aromatic Compounds ▶ List of Issues ▶ Volume 42, Issue 7
▶ Green Synthesis, Molecular Docking, In S

Polycyclic Aromatic Compounds >

Volume 42, 2022 - Issue 7


197 | 4

0

Views | CrossRef citations to date | Altmetric

Research Articles

Green Synthesis, Molecular Docking, *In Silico* ADME and Biological Evaluation of Methoxy Substituted 1,5-Benzodiazepines as Potential Antioxidant, Anti-Inflammatory, and Antidiabetic Agents

Nargisbano A. Peerzade, Shravan Y. Jadhav , Bhushan D. Varpe, Amol A. Kulkarni & Raghunath B. Bhosale

Pages 3939-3952 | Received 23 Mar 2020, Accepted 12 Jan 2021, Published online: 08 Feb 2021

 Cite this article  <https://doi.org/10.1080/10406638.2021.1878244>



 Full Article

 Figures & data

 References

 Citations

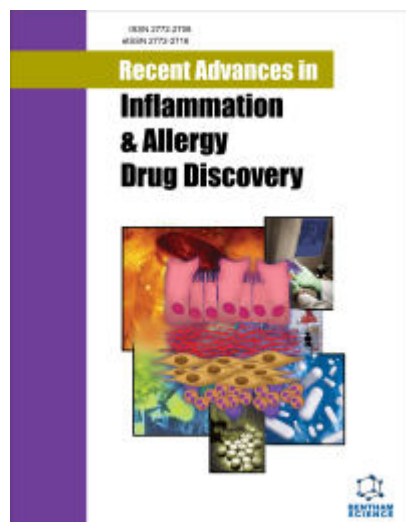
 Metrics

 Reprints & Permissions

Read this article

Abstract

Present work involves the green and environmentally benign synthesis of 1,5-benzodiazepines using iodine as a catalyst and PEG-400 as a green solvent at room temperature that gives an excellent yield of products. All the synthesized compounds were screened for their anti-inflammatory, antioxidant, and antidiabetic activity. All synthesized compounds showed excellent antioxidant activity against DPPH and



Syntheses, Molecular Docking and Biological Evaluation of 2-(2- hydrazinyl)thiazoles as Potential Antioxidant, Anti-Inflammatory and Significant Anticancer Agents

Buy Article:

\$68.00 + tax

(Refund Policy)

ADD TO CART

BUY NOW

Authors: Raut, Dattatraya G.; Bhosale, Raghunath B.; Lawand, Anjana S.; Hublikar, Mahesh G.; Kadu, Vikas D.; Patil, Sandeep B.; Choudhari, Prafulla B.

Source: Recent Advances in Inflammation & Allergy Drug Discovery, Volume 16, Number 2, 2022, pp. 96-106(11)

Publisher: Bentham Science Publishers

DOI: <https://doi.org/10.2174/2772270816666220902094019>

...
Abstract

References

”
Citations

≡
Supplementary Data

Background: Recently, researchers have worked on the development of new methods for the synthesis of bioactive heterocycles using polyethylene glycol as a green solvent. In this context, we report the synthesized 2-(2-hydrazinyl) thiazoles for their in vitro antioxidant, in vitro anti-inflammatory and in vitro anti-cancer activities.

Objective: The objective of the study was to develop novel antioxidant, anti-inflammatory and anti-cancer drugs.

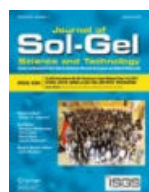
Methods: At the outset, the condensation of substituted acetophenones 1, thiosemicarbazide 2, and α -haloketones 3 was carried out using PEG-400 (20 mL) in the presence of 5 mol% glacial acetic acid to afford thiosemicarbazones intermediate. Furthermore, these thiosemicarbazones were reacted with α -haloketones 3 to obtain appropriate 2-(2-hydrazinyl) thiazoles. The synthesized compounds were in vitro tested for their antioxidant, anti-inflammatory, and anti-cancer activity.


[Home](#) [Journal of Sol–Gel Science and Technology](#) [Article](#)

Effect of embedding aluminium and yttrium on the magneto–optic properties of lanthanum spinel ferrite nanoparticles synthesised for photocatalytic degradation of methyl red

Original Paper: Nano-structured materials (particles, fibres, colloids, composites, etc.)

Published: 08 October 2022

Volume 104, pages 354–364, (2022) [Cite this article](#)[Journal of Sol–Gel Science and Technology](#)[Aims and scope](#)[Submit manuscript](#)

[Subiya K. Kazi](#), [Radhakrishnan M. Tigote](#) , [Vandana A. Gaikwad](#), [Dhanraj P. Kamble](#), [Pravin S. Bhale](#), [Sadanand N. Shringare](#), [Pramod G. Musrif](#) & [Shaukatali N. Inamdar](#)

 257 Accesses  4 Citations [Explore all metrics](#) →

Abstract

The sol–gel approach was used to synthesise lanthanum and aluminium doped yttrium ferrite nanoparticles. The absorption peak is observed at ~265 nm corresponds to band gap of 2.9–3.1 eV. X-ray diffraction (XRD), field emission scanning electron microscopy (FE–SEM) and High–Resolution Transmission Electron Microscopy (HR – TEM) were used to analyse the structural and microstructural properties of the product. The particle size calculated to be 60–110 nm from the HR–TEM analysis. The vibrating sample



3-Substituted-2-oxindole derivatives: Design, synthesis and their anti-tuberculosis and radical scavenging dual-action studies

Mahesh Hublikar^a, Vikas Kadu^a, Dattatraya Raut^a, Sachin Shirame^a, Sivaraj Anbarasu^b, Muhanna K. Al-Muhanna^c, Parameshwar Makam^{d,e}  , Raghunath Bhosale^a 

[Show more](#) 

 Share  Cite

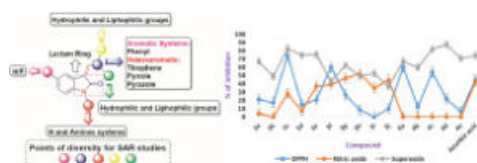
<https://doi.org/10.1016/j.molstruc.2022.132903> 

[Get rights and content](#) 

Abstract

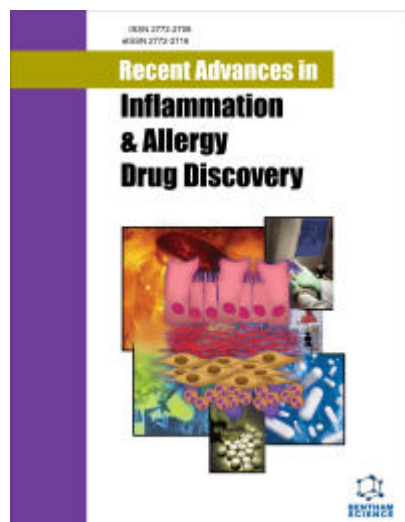
Despite the fact that the oxindole has long been thought to be the best structural scaffold, research shows that oxindole derivatives are being studied the least against tuberculosis (TB), a historically airborne illness. The present research focuses on the development of 3-substituted-2-oxindole derivatives with a combination of structural and functional alterations in order to create dual-active new chemical entities with anti-TB, anti-oxidant, and radical-scavenging properties. The physicochemical properties of these compounds are in line with the theories of Lipinski, Veeber, and Leeson. SWISSADME was used to assess the drug-like molecular nature and pharmacokinetics features. Multistep synthetic procedures were used to synthesise the chemicals. Based on *in vitro* anti-TB and radical scavenging property analyses, the synthesised 3-substituted-2-oxindole derivatives were confirmed as dual-active compounds. The synthesised 3-substituted-2-oxindole derivatives were confirmed to possess the dual therapeutics action. The discovery of radical scavengers **4d**, **4a**, **4c**, **3d**, **3c**, **3f**, **3e**, **3h**, **3g**, and **3j** that are more potent than Ascorbic acid, the standard. The molecules, **3c**, **3j**, **3b**, and **4b** have exhibited moderate to active anti-TB properties with a % reduction values of 46.42%, 40.89%, 39.76% and 39.32%, respectively. The findings suggest that fine-tuning molecules might result in compounds with better anti-TB and radical-scavenging properties.

Graphical abstract



[Download : Download high-res image \(117KB\)](#)

[Download : Download full-size image](#)



Syntheses, Molecular Docking and Biological Evaluation of 2-(2- hydrazinyl)thiazoles as Potential Antioxidant, Anti-Inflammatory and Significant Anticancer Agents

Buy Article:

\$68.00 + tax

(Refund Policy)

[ADD TO CART](#)

[BUY NOW](#)

Authors: Raut, Dattatraya G.; Bhosale, Raghunath B.; Lawand, Anjana S.; Hublikar, Mahesh G.; Kadu, Vikas D.; Patil, Sandeep B.; Choudhari, Prafulla B.

Source: Recent Advances in Inflammation & Allergy Drug Discovery, Volume 16, Number 2, 2022, pp. 96-106(11)

Publisher: Bentham Science Publishers

DOI: <https://doi.org/10.2174/2772270816666220902094019>



Abstract



References



Citations



Supplementary Data

Background: Recently, researchers have worked on the development of new methods for the synthesis of bioactive heterocycles using polyethylene glycol as a green solvent. In this context, we report the synthesized 2-(2-hydrazinyl) thiazoles for their in vitro antioxidant, in vitro anti-inflammatory and in vitro anti-cancer activities.

Objective: The objective of the study was to develop novel antioxidant, anti-inflammatory and anti-cancer drugs.

Methods: At the outset, the condensation of substituted acetophenones 1, thiosemicarbazide 2, and α -haloketones 3 was carried out using PEG-400 (20 mL) in the presence of 5 mol% glacial acetic acid to afford thiosemicarbazones intermediate. Furthermore, these thiosemicarbazones were reacted with α -haloketones 3 to obtain appropriate 2-(2-hydrazinyl) thiazoles. The synthesized compounds were in vitro tested for their antioxidant, anti-inflammatory, and anti-cancer activity.

[Home](#) [Polymer Bulletin](#) [Article](#)

Room-temperature ammonia gas sensor based on carboxylic acid-doped polyaniline

Original Paper Published: 04 April 2022

Volume 80, pages 3183–3195, (2023) [Cite this article](#)[Polymer Bulletin](#)[Aims and scope](#)[Submit manuscript](#)

[Renukacharya G. Khanapure](#), [Anil A. Ghanwat](#), [Sharad K. Awate](#), [Uttam S. Gawali](#), [Rajesh J. Kavade](#), [Pravin H. Salunkhe](#) & [Suresh V. Patil](#)



366 Accesses 2 Citations [Explore all metrics](#) →

Abstract

Polyaniline films modified by carboxylic acid dopents have been formed by in situ chemical oxidative polymerization using ammonium persulfate as an oxidizing agent and investigated for NH_3 gas sensor. We demonstrate that butyric acid-doped polyaniline (BPANI) film is highly selective to NH_3 with high selectivity (513% at 100 ppm), fast response time (8.94 s). Furthermore, good reproducibility was also observed at 100 ppm. The room-temperature functioning of the sensor is critical, which facilitate low-power operation and also enhances the life time of the sensor. The results indicate that the carboxylic acid-doped polyaniline films are promising for NH_3 gas detection.



High-performance supercapacitive polyazomethines: Room temperature synthesis and their characterizations

Amburaya S. Birajdar^a, Shailesh G. Pawar^b, Anil A. Ghanwat^c, Vijaykumar P. Ubale^a  

Show more 

 Share  Cite

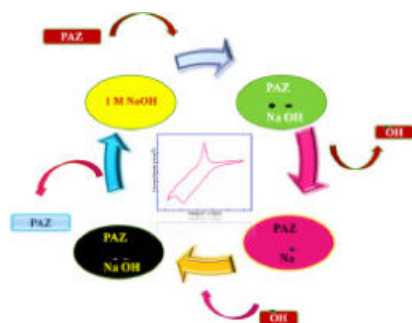
<https://doi.org/10.1016/j.molstruc.2023.137173> 

[Get rights and content](#) 

Abstract

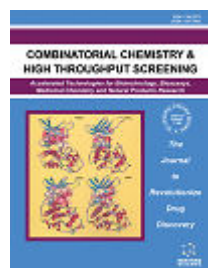
A room temperature synthesized polyazomethine (PAZ) series via polycondensation route comprising monomer as 5-[4-(5-formylfuran-2-yl)phenyl]furan-2-carbaldehyde (III) and various diamines to engineer PAZ-1 to PAZ-5 polymers. Physico-chemical properties of as-prepared PAZ series have confirmed by X-ray diffraction (XRD), field emission scanning electron microscopy (FESEM), energy dispersive X-ray analysis (EDAX), transmission electron microscopy (TEM) and spectroscopic techniques. In FTIR, --CH=N-- peak at 1600 cm^{-1} attributed to the successful conversion of dicarbaldehyde (III) and diamines into polyazomethine. The amorphous nature of PAZ series is depicted by XRD diffractograph. The marigold like morphology of polyazomethines has revealed in FESEM and TEM images. The chemical composition and elemental distribution on PAZ surface has mapped using EDAX technique. The electrochemical activity of as-prepared polyazomethine electrodes has investigated by cyclic voltammetry (CV), galvanostatic charge discharge (GCD) and electrochemical impedance spectroscopy (EIS) techniques. The supercapacitive performance of PAZ-1 to PAZ-5 electrodes in various electrolytes as NaOH, KOH, Na_2SO_4 , and LiCl by varying concentrations, the PAZ-2 electrode exhibits elevated specific capacitance (C_p) of 831 Fg^{-1} , specific energy (SE) 22.99 Whrk^{-1} , specific power (SP) 82.87 KWkg^{-1} in optimum 1 M NaOH electrolyte. The distinguished PAZ-2 electrode after successive 500 cycles has delivered exceptional stability, 71.65% in 1 M NaOH.

Graphical Abstract



Download : [Download high-res image \(118KB\)](#)

Download : [Download full-size image](#)



Synthesis of Novel Hydrazones of Levofloxacin Related Molecule and their In Vitro Evaluation as Antioxidant, and Molecular Docking Studies

Buy Article:

\$68.00 + tax

(Refund Policy)

ADD TO CART

BUY NOW

Authors: Kashid, Bharat B.; Kilbale, Jaydeo T.; Wani, Kishor D.; Pawar, Suhas. M.; Khedkar, Vijay M.; Ghanwat, Anil A.

Source: Combinatorial Chemistry & High Throughput Screening, Volume 25, Number 2, 2022, pp. 274-283(10)

Publisher: Bentham Science Publishers

DOI: <https://doi.org/10.2174/1386207323666201229150734>



...
Abstract

References

Citations

Supplementary Data

Objective: The research work aims to synthesize novel series of hydrazones and antioxidant screening. It also aims to evaluate the binding affinities and in silico methods for identifying possible drug targets of synthesized compounds.





Methods: This report briefly explains the synthesis of a novel series of hydrazones. It was synthesized via hydrazinolysis of esters to obtain hydrazide, treated with aldehyde and acetophenone to get hydrazones. The spectral confirmed hydrazones exhibited excellent to comparable anti-oxidant as compared to the standard drugs Butylated hydroxytoluene (BHT) and Ascorbic acid. Molecular docking on myeloperoxidase (MPO) demonstrated the ability of this scaffold to correctly recognize the target and engage in significant bonded and non-bonded interactions with key residues therein.

Results and Discussion: In this study, we report effectively synthesized compounds BK-35, BK- 41, BK-26, BK-28, and BK-39 that showed the best DPPH radical scavenging activity. The docking results clearly showed the binding mode of hydrazones into the active site of Myeloperoxidase (MPO). In in-silico results, none of the synthesized compounds, BK-24 to BK- 41, violated Lipinski's rule of five ($\text{miLog } P \leq 5$).

Conclusions: In vitro preliminary anti-oxidant screening results in support by in Silico binding affinity data of novel hydrazones of levofloxacin related molecules BK-24 to BK-41 reported here have emerged as excellent anti-oxidant agents. The inference derived from the in vitro anti-oxidant screening data and the quantitative insights derived from the per-residue interaction analysis with MPO enzyme are now being fruitfully utilized for site-specific mutation around the nucleus to identify selective and potent anti-oxidants.



Development of new efficient and cost effective liquid-liquid extractive determination method for cobalt(II): Analysis of water, alloys and nano powder

Ashwini V. Sadlapurkar^{a b}, Umesh B. Barache^{b c}  , Abdul B. Shaikh^b, Anjana S. Lawand^c,
Shashikant H. Gaikwad^b  , Tukaram N. Lokhande^b

[Show more](#) 

 Outline |  Share  Cite

<https://doi.org/10.1016/j.jtemin.2022.100026> 

[Get rights and content](#) 

Under a Creative Commons [license](#) 

[open access](#)

Abstract

Background

The renowned biological role of cobalt is it is main component of vitamin B12, however other cobalt compounds have been listed as toxic for the environment as well as to human.

Methods

Various samples from different sources are analyzed for their cobalt(II) content by extraction followed by UV-visible spectrophotometry and compared with atomic absorption spectrophotometry.

Results

In this article, the chromogenic reagent 2-chlorobenzaldehyde thiocarbohydrazone is introduced for extractive spectrophotometric determination of cobalt(II) from various samples. This reagent forms yellow colored 1:2:2 [Co(II)–2CBTCH–iodide] complex in dichloromethane which was extracted from an acetate buffer having pH of 4.2 in presence of potassium iodide solution which was stable up to 48h. The absorbance of the complex exhibit peak absorbance at 400nm. The present technique was optimized for numerous influences and the interference of other ion has also been cautiously studied. The calculated values of molar absorptivity and Sandell's sensitivity of the complex are found to be $0.3006 \times 10^4 \text{ mol}^{-1} \text{ cm}^{-1}$ and $0.0196 \mu\text{g cm}^{-2}$ respectively. The technique conforms Beer's law up to $13 \mu\text{g mL}^{-1}$ with 0.999 correlation coefficient of the [Co(II)–2CBTCH–iodide] complex, which specifies linearity between the two variables. For five replicate determinations ($n=5$), the relative standard deviation was 1.18 with the regression equations as $y=0.0672 x+0.01$ with $R^2=0.999$ as the correlation coefficient. The recovery percentages were warranted the accuracy and found around 99.0%.

Conclusion

In Vitro Anticancer Screening, Molecular Docking and Antimicrobial Studies of MBPP-Triazole-Based Nickel(II) Metal Complexes

by Conclusions

Supplementary

1, 2 Materials

Author

Contributions

3 Funding

Pratibha C. Dhale (https://sciprofiles.com/profile/author/UjBsYWR2RVMycVUxbllrd2o4K0cwL05uaUtzQjdIZlgvT0NGbjVXUk1Gcz0

2 Institutional

Review

2 Board

Statement

Chandan Shivamallu (https://sciprofiles.com/profile/359968?utm_source=mdpi.com&utm_medium=website&utm_campaign=avatar_

4 Informed

Consent

Statement

5 Data

Availability

6 Statement

Acknowledgments

6 Conflicts

6, 7 Interest

8 Author

8 Contributions

9 Funding

9 Institutional

9 Review

9 Board

9 Statement

9 Informed

9 Consent

9 Statement

9 Data

9 Availability

9 Statement

9 Acknowledgments

9 Conflicts

9 Interest

9 Author

9 Contributions

9 Funding

9 Institutional

9 Review

9 Board

9 Statement

9 Informed

9 Consent

9 Statement

9 Data

9 Availability

9 Statement

9 Acknowledgments

9 Conflicts

9 Interest

9 Author

9 Contributions

9 Funding

9 Institutional

9 Review

9 Board

9 Statement

9 Informed

9 Consent

9 Statement

9 Data

9 Availability

9 Statement

View PDF (chrome-extension://dagcmkpagjlhakfdhnbomgmjdpkdklff/enhanced-reader.html?openApp&pdf=https%3A%2F%2Fmdpi-

1 Arts, Science and Commerce College, Naldurg, Osmanabad 413 602, Maharashtra, India

2 Chemistry Research Laboratory, Department of Chemistry, Shri Shri Mahavidyalaya, Solapur 413 411, Maharashtra, India

3 School of Chemical Sciences, Poojashlok Ahilyadevi Holkar Solapur University, Solapur 413 255, Maharashtra, India

4 Department of Biotechnology and Bioinformatics, School of Life Sciences, JSS Academy of Higher Education and Research, Mysuru 570 015, Karnataka, India

5 Department of Chemistry, N. K. Orchid College of Engineering and Technology, Solapur 413 002, Maharashtra, India

6 Biology Department, Faculty of Sciences, King Khalid University, Abha 61421, Saudi Arabia

7 Cell Culture Lab, Egyptian Organization for Biological Products and Vaccines (VACSERA Holding Company), 51 Wezaret El-Zeraa St., Agouza, Giza 12311, Egypt

8 Division of Biochemistry, School of Life Sciences, JSS Academy of Higher Education and Research, Mysuru 570 015, Karnataka, India

9 Department of Hospital Surgery, N.I. Pirogov Russian National Research Medical University (RNRMU), 117997 Moscow, Russia

10 Departament de Química, Facultat de Ciències, Universitat de les Illes Balears, E-07122 Palma de Mallorca, Spain

+ Show full affiliation list

* Authors to whom correspondence should be addressed.

Molecules **2022**, *27*(19), 6548; <https://doi.org/10.3390/molecules27196548> (<https://doi.org/10.3390/molecules27196548>)

Submission received: 8 September 2022 / Revised: 27 September 2022 / Accepted: 29 September 2022 / Published: 3 October 2022





(This article belongs to the Special Issue **Chemistry of Nitrogen Heterocyclic Compounds** (/journal/molecules/special_issues/nitrogen_heterocycl_compd))



Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy

Volume 285, 15 January 2023, 121918

Analytical optimization of liquid–liquid extractive spectrophotometric assessment protocol for tetravalent platinum: Analysis of environmental samples and cisplatin

Abdul B. Shaikh^a, Umesh B. Barache^{a, b}  , Anjana S. Lawand^b, Ganesh S. Kamble^{c, d}, Muddsar L. Gaur^e, Shashikant H. Gaikwad^a  

[Show more](#) 

 Share  Cite

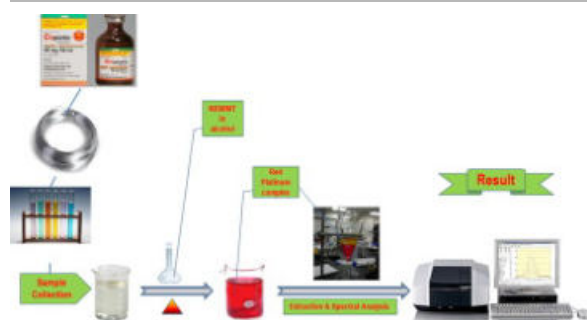
<https://doi.org/10.1016/j.saa.2022.121918> 

[Get rights and content](#) 

Abstract

An easy and reliable method is optimized for extractive spectrophotometric assessment of platinum(IV) with 4-(4'-nitrobenzylideneimino)-3-methyl-5-mercapto-1,2,4-triazole as an extractant. The basis of this method is the formation of red platinum(IV) complex with the above reagent in acetate buffer medium (pH 5.0) and extraction in chloroform. Good linearity with regression equation as $y = 1.011 \times 10^4 x + 0.002$ having correlation coefficient (R^2) of 0.998 over concentration up to $17.5 \mu\text{g mL}^{-1}$ of platinum(IV) was achieved with apparent molar absorptivity of $1.011 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$. The limit of detection ($0.22 \mu\text{g mL}^{-1}$), limit of quantification ($0.73 \mu\text{g mL}^{-1}$) and Sandell's sensitivity ($0.0193 \mu\text{g cm}^{-2}$) were also estimated. The interference of various cations was removed by using proper masking agents and consequently by using EDTA and citrate to mask certain transition metals, the method becomes highly specific for platinum(IV), including the effects of platinum group metals. The method is effectively used for determination of platinum(IV) from environmental and real samples such as alloys, catalysts, thermocouple wire and pharmaceutical sample.





Graphical abstract



[Download](#) : [Download high-res image \(112KB\)](#)



Potential arsenic–chromium–lead Co-contamination in the hilly terrain of Arunachal Pradesh, north-eastern India: Genesis and health perspective

Ritusmita Goswami^a  , Chandrashekhar Bhagat^{b f}, Igo Lollen^c, Nikita Neog^a, Umesh B. Barache^d, Ritu Thakur^e, Jorgen Mahlkecht^g, Manish Kumar^{f g}  

[Show more](#) 

 Share  Cite

<https://doi.org/10.1016/j.chemosphere.2023.138067> 

[Get rights and content](#) 

Abstract

In the recent times, multi-metal co-contamination in the groundwater of various parts of the globe has emerged as a challenging environmental health problems. While arsenic (As) has been reported with high fluoride and at times with uranium; and Cr & Pb are also found in aquifers under high anthropogenic impacts. The present work probably for the first time traces the As–Cr–Pb co-contamination in the pristine aquifers of a hilly terrain that are under relatively less stress from the anthropogenic activities. Based on the analyses of twenty-two (n=22) groundwater (GW) samples and six (n=6) sediment samples, it was found that Cr being leached from the natural sources as evident from 100% of samples with dissolve Cr exceeding the prescribed drinking water limit. Generic plots suggests rock-water interaction as the major hydrogeological processes with mixed $\text{Ca}^{2+}\text{-Na}^+\text{-HCO}_3^-$ type water. Wide range of pH suggests localized human interferences, as well as indicative of both calcite and silicate weathering processes. In general water samples were found high only with Cr and Fe, however all sediment samples were found to contain As–Cr–Pb. This implies that the groundwater is under-risk of co-contamination of highly toxic trio of As–Cr–Pb. Multivariate analyses indicate that the changing pH as the causative factor for Cr leaching into the groundwater. This is a new finding for a pristine hilly aquifers, and we suspect such condition may also be present in other parts of globe, and thus precautionary investigations are needed to prevent this catastrophic situation to arise, and to alert the community in advance.

Graphical abstract



Print ISSN : 2277 - 7830 Online ISSN : 2348 - 5922 indexed with J Gate, i scholar and Google scholar

IBMRD's Journal of Management & Research

Biannual Peer reviewed Blind fold Journal



[HOME](#) | [LOGIN](#) | [CURRENT](#) | [ARCHIVES](#) | [EDITORIAL BOARD](#) | [AUTHOR INDEX](#) | [CALL FOR PAPERS](#) | [PLAGIARISM POLICY](#)

Home > Volume 11, Issue 2, September 2022 > Raut

Total views : 288

Cyber Security Awareness: A Movement of Digital Literacy Towards making of Digital India

Dr. Shriram D. Raut ^{1*}, Dr. Ashok R. Shinde ², Dr. M. K. Patil ³

Affiliations

1. School of Computational Sciences, P. A. Holkar Solapur University, Solapur, India
2. School of Computational Sciences, P. A. Holkar Solapur University, Solapur
3. Bharati Vidyapeeth, AKIMSS, Solapur, India

DOI: 10.17697/ibmrd/2022/v11i2/172617

| | |
|--|-----------------------------------|
| | Our Subscribers |
| | Aim & Scope |
| | Submit an Article |
| | Advanced Search |
| | Notification |
| | Title Index |
| | Announcements |
| | Register |
| | Subscription |

Punyashlok Ahilyadevi Holkar Solapur University

Solapur Pune National Highway

Kegaon

Solapur

Consultancy Charges School of Allied Health Sciences

Ledger Account

1-Apr-2022 to 31-Mar-2023

| Date | Particulars | Vch Type | Vch No. | Debit | Page 1 Credit |
|-----------|---|-------------|---------|-----------|------------------|
| 5-8-2022 | Dr Central Bank of India NEFT A/c No. 3177057679 UTR No - BD22071278613618 Dt - 12/07 /2022 | Receipt | 8589 | | 10,000.00 |
| 10-8-2022 | Cr (as per details) | Journal | 923 | 10,000.00 | |
| | Kane Manali -Payable | 2,500.00 Cr | | | |
| | Shri Gardi Chandrakant Gopal -Payable | 2,500.00 Cr | | | |
| | Ghorpade Sujay Annaso -Payable | 2,500.00 Cr | | | |
| | Dr.Jagtap Abhijit Hanumant -Payable | 2,500.00 Cr | | | |
| | Being Amount Credited Against Consultancy Charges | | | | |
| | | | | 10,000.00 | 10,000.00 |

[illegible]

| PAHSUS Summer Internship Program Attendance Nutrition & Dietetics- FIRST WEEK | | | | | | | |
|---|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|
| | | 23rd May | 24th May | 25th May | 26th May | 27th May | 28th May |
| | Student's Name | Monday | Tuesday | Wednesday | Thursday | Friday | Saturday |
| Morning Schedule | Prajakta Madake | <u>Mandak</u> | <u>Mandak</u> | <u>Mandak</u> | <u>Mandak</u> | <u>Mandak</u> | <u>Mandak</u> |
| | Sakshi Karale | <u>sumk.</u> | <u>sumk.</u> | <u>sumk.</u> | <u>sumk.</u> | <u>sumk.</u> | <u>sumk.</u> |
| | Sneha Shingade | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> |
| | Vaishnavi Gund | <u>Vaishu</u> | <u>Vaishu</u> | <u>Vaishu</u> | <u>Vaishu</u> | <u>Vaishu</u> | <u>Vaishu</u> |
| | Pawan Gunnal | <u>Gunn</u> | <u>Gunn</u> | <u>Gunn</u> | <u>Gunn</u> | <u>Gunn</u> | <u>Gunn</u> |
| Sign of Head | | | | | | | |
| Afternoon Session | Prajakta Madake | <u>Mandak</u> | <u>Mandak</u> | <u>Mandak</u> | <u>Mandak</u> | <u>Mandak</u> | <u>Mandak</u> |
| | Sakshi Karale | <u>sumk.</u> | <u>sumk.</u> | <u>sumk.</u> | <u>sumk.</u> | <u>sumk.</u> | <u>sumk.</u> |
| | Sneha Shingade | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> |
| | Vaishnavi Gund | <u>Vaishu</u> | <u>Vaishu</u> | <u>Vaishu</u> | <u>Vaishu</u> | <u>Vaishu</u> | <u>Vaishu</u> |
| | Pawan Gunnal | <u>Gunn</u> | <u>Gunn</u> | <u>Gunn</u> | <u>Gunn</u> | <u>Gunn</u> | <u>Gunn</u> |
| Sign of Head | | | | | | | |

verified.
 18/12
Medical Officer
Health Centre
Punyashlok Ahilyadevi Holkar
Solapur University, Solapur.



| PAHSUS Summer Internship Program Attendance Nutrition & Dietetics- SECOND WEEK | | | | | | | |
|--|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|
| | | 30th May | 31st May | 1st June | 2nd June | 3rd June | 4th June |
| | Student's Name | Monday | Tuesday | Wednesday | Thursday | Friday | Saturday |
| Morning Schedule | Prajakta Madake | <u>Mandake</u> | <u>Mandake</u> | <u>Mandake</u> | <u>Mandake</u> | <u>Mandake</u> | <u>Mandake</u> |
| | Sakshi Karale | <u>Karale</u> | <u>Karale</u> | <u>Karale</u> | <u>Karale</u> | <u>Karale</u> | <u>Karale</u> |
| | Sneha Shingade | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> |
| | Vaishnavi Gund | <u>Gund</u> | <u>Gund</u> | <u>Gund</u> | <u>Gund</u> | <u>Gund</u> | <u>Gund</u> |
| | Pavan Gunnal | <u>Gunnal</u> | <u>Gunnal</u> | <u>Gunnal</u> | <u>Gunnal</u> | <u>Gunnal</u> | <u>Gunnal</u> |
| Sign of Head | | | | | | | |
| Afternoon Session | Prajakta Madake | <u>Mandake</u> | <u>Mandake</u> | <u>Mandake</u> | <u>Mandake</u> | <u>Mandake</u> | <u>Mandake</u> |
| | Sakshi Karale | <u>Karale</u> | <u>Karale</u> | <u>Karale</u> | <u>Karale</u> | <u>Karale</u> | <u>Karale</u> |
| | Sneha Shingade | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> | <u>Shingade</u> |
| | Vaishnavi Gund | <u>Gund</u> | <u>Gund</u> | <u>Gund</u> | <u>Gund</u> | <u>Gund</u> | <u>Gund</u> |
| | Pavan Gunnal | <u>Gunnal</u> | <u>Gunnal</u> | <u>Gunnal</u> | <u>Gunnal</u> | <u>Gunnal</u> | <u>Gunnal</u> |
| Sign of Head | | | | | | | |

5th JUNE

SUNDAY

Mandake

Karale

Shingade

Gund

Gunnal

Mandake

Karale

Shingade

Gund

Gunnal

Mandake

Karale

Shingade

Gund

Gunnal


Verified
12/6
Medical Officer
Health Centre
Punyashlok Ahilyadevi Holkar
Solapur University, Solapur.



| CERTIFICATE COURSE IN ACUPRESSURE | | | |
|-----------------------------------|-----------------------------------|---------|-------------------------------|
| Sr. No. | Course Name | FormNo | Student Name |
| 1 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-01 | MANE AMAR ASHOK |
| 2 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-02 | JOSHI ASHWINI DURGAPRASAD |
| 3 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-03 | MAYNALE SACHIN IRANNA |
| 4 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-04 | KASBE RAJRATNA JAGJERAO |
| 5 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-05 | DUPARGUDE VIJAYALAXMI NAGNATH |
| 6 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-06 | SHAH KAUSHIK MAHENDRABHAI |
| 7 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-07 | DILPAK AJIT SURESH |
| 8 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-08 | MASKE SAMBHAJI SIDRAM |
| 9 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-09 | DUPARGUDE PRASNNAJIT NAGNATH |
| 10 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-10 | EKHANDE SANGEETA PRASHANT |
| 11 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-12 | KATOTE RAJENDRA SHRINIVAS |
| 12 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-13 | MUJAWAR IBRAHIM MAHIBUB |
| 13 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-14 | RATHOD SANKET BHARAT |
| 14 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-15 | ALAGUNDGI SHRIKANT MALAKAPPA |
| 15 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-16 | SHAIKH SULEMAN SIRAJAHMED |
| 16 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-17 | HAMBIRE SHRINIWAS DAYANANDRAO |
| 17 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-18 | TIKE SACHIN GOVINDRAO |
| 18 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-19 | GHADGE SWATI GOPALRAO |
| 19 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-20 | GAIKWAD RAJNI NANASAHEB |
| 20 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-21 | RATHI RAJESH SHIVNARAYAN |
| 21 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-22 | UDANSHIV SHRUTIKA MOHAN |
| 22 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-23 | SHAIKH SAIMAAMBAR NAIMODDIN |
| 23 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-24 | SHAIKH SAIF ABDUL SALAM |
| 24 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-25 | BHOSALE DIGVIJAY KALIDAS |
| 25 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-27 | PATIL RUPALI SIDRAM |
| 26 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-28 | ADAM GANESH KRISHNAHARI |
| 27 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-29 | BHORE MANISHA SANTOSH |
| 28 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-30 | JOSHI ARTI CHINTAN |
| 29 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-31 | MARU DHANASHRI NITEN |
| 30 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-32 | SALE SHRIMANT CHANDRU |
| 31 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-33 | PAWAR PORNIMA SANJAY |
| 32 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-34 | AIWALE KARISHMA ABA |
| 33 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-35 | POLANI DARSHANA NARESH |
| 34 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-36 | SATHE ANITA RAMCHANDRA |
| 35 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-37 | ALANDKAR CHETANA ANIL |
| 36 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-38 | PATIL DHARMRAJ VASANTRAO |
| 37 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-41 | KULKARNI SURYAKANT SUDHAKAR |
| 38 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-42 | ALANDKAR ANIL BHALCHANDRA |
| 39 | CERTIFICATE COURSE IN ACUPRESSURE | AP22-43 | GURAV SATAPPA RAJARAM |



18/12/20
Co-ordinator
 School of Allied Health Science
 Punyashlok Ahilyadevi Holkar
 Solapur University, Solapur


Director
 School of Allied Health Science
 Punyashlok Ahilyadevi Holkar
 Solapur University, Solapur