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M.Sc. Physics (Solid State) (Semester - I) (New) (NEP CBCS)
Examination: March/April – 2026
Mathematical Physics (2307101)

Day & Date: Friday, 17-04-2026
 Time: 03:00 PM To 05:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
 2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative. (MCQ)

08

- 1) Function $f(Z) = \frac{1}{z}$ has a _____.
 - a) Branch point
 - b) Simple pole at $z = 0$
 - c) Essential singularity
 - d) Removable singularity
- 2) The integral of $1/Z$ along the circle $|Z| = 1$ is _____.
 - a) 0
 - b) $i\pi$
 - c) $2\pi i$
 - d) $-2\pi i$
- 3) The dimensionality of vector space spanned by $\{(1,0,0), (0,1,0), (0,0,1)\}$ _____.
 - a) 1
 - b) 2
 - c) 3
 - d) 4
- 4) The dimensionality of all 2×2 matrices is _____.
 - a) 4
 - b) 9
 - c) 12
 - d) 3
- 5) The general solution of $\frac{dy}{dx} = ky$ is _____.
 - a) $y = C e^{kx}$
 - b) $y = kx + C$
 - c) $y = C x^k$
 - d) $y = c + kx$
- 6) If the roots of characteristic equation are complex, then the solution is _____.
 - a) Real and distinct
 - b) Real and repeated
 - c) Complex conjugate
 - d) Imaginary
- 7) The Fourier series of a square wave contains _____.
 - a) Only sine terms
 - b) Only cosine terms
 - c) Both sine and cosine
 - d) Constant terms only

- 8) The coefficient a_n in a Fourier series represents _____.
- Average (DC) value of the function
 - Maximum amplitude
 - Frequency
 - Phase angle

B) State True or False.**04**

- Analytic functions automatically satisfy Laplace's equation and are therefore harmonic.
- Parseval's theorem relates the total energy (or power) of a signal in the time domain to that in the frequency domain.
- Linearly dependent vectors can form a basis for a vector space.
- A second-order homogeneous differential equation with constant coefficients always has exponential-type solutions.

Q.2 Answer the following. (Any Six)**12**

- a) Find the complex conjugate of the following equations.

i) $Z_1 = -4 + 7i$

ii) $Z_2 = 3 - 6i$

- b) Find the cube root of the following number $Z = 8i$

- c) Find the general solution of the given differential equation.

$$\frac{d^2y}{dt^2} + 5\frac{dy}{dt} + 6y = 12e^{-2t}$$

- d) Find the general solution of the given differential equation

$$\frac{d^2y}{dt^2} - 9y = 0$$

- e) Determine whether the vectors $V_1 = (1,0,2)$ and $V_2 = (0,1,3)$ are linearly independent.

- f) State Fourier theorem and briefly explain its significance in signal processing.

- g) Define the Laplace transform of a function $f(t)$ and explain its significance.

- h) Find the inverse of the matrix $A = \begin{bmatrix} 3 & 2 \\ 1 & 4 \end{bmatrix}$

Q.3 Answer the following. (Any Three)**12**

- a) State and Explain Cauchy's theorem in detail.

- b) Let $A = \begin{bmatrix} 1 & 0 \\ 0 & 0 \end{bmatrix}$, show $A^2 = A$ but $I - A = 0$ is invertible.

- c) Find the Laplace transform of $f(t) = te^{-2t}$ for $t \geq 0$.

- d) State and prove superposition principle.

Q.4 Answer the following. (Any Two)

12

- a) Find the general solution of $y'' - y' - 6y = 3e^{2x}$
- b) Evaluate the integral $f(z) = \oint \frac{\cos(z)}{z^2+1} dz$
- c) Find the eigen values and eigen vectors of the matrix

$$A = \begin{vmatrix} 3 & -1 & 0 \\ -1 & 3 & -1 \\ 0 & -1 & 3 \end{vmatrix}$$

Q.5 Answer the following. (Any Two)

12

- a) Evaluate the integral $f(z) = \oint \frac{\sin(z)}{z^2+4} dz$ where, C is the circle $|Z| = 3$ traversed counter clockwise.
- b) Compute the Fourier series of the function $f(x) = x^2$ on the interval $-\pi \leq x \leq \pi$ period 2π
- c) Find the inverse of the matrix $B = \begin{vmatrix} 1 & 2 & 3 \\ 0 & 1 & 4 \\ 5 & 6 & 0 \end{vmatrix}$ and verify $B \times B^{-1} = I$

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Set **P**

M.Sc. Physics (Solid State) (Semester - I) (New) (NEP CBCS)
Examination: March/April - 2026
Solid State Physics (2307102)

Day & Date: Monday, 20-04-2026
 Time: 03:00 PM To 05:30 PM

Max. Marks: 60

- Instructions:** 1) All questions are compulsory.
 2) Figures to the right indicate full marks.
 3) Draw neat labelled diagrams wherever necessary.

Q.1 A) Choose correct alternative. (MCQ)**08**

- 1) Which of the following statement is wrong in case of a free electron?
 - a) A free electron is not attached to atom or molecule
 - b) A free electron contributes for conduction
 - c) A free electron responds to external fields.
 - d) A free electron is always bound.

- 2) The Bohr magneton, $\mu_B =$ _____.

a) $\frac{e\hbar}{2m_e}$	b) $\frac{e\hbar}{2\pi m_e}$
c) $\frac{eh}{2m_e}$	d) $\frac{e\hbar}{4\pi m_e}$

- 3) Susceptibility of diamagnetic materials is _____.
 - a) a small negative quantity
 - b) small positive quantity
 - c) large positive quantity
 - d) large negative quantity

- 4) The smallest repeating unit of the lattice is called _____.

a) unit cell	b) primitive
c) nonprimitive	d) centrosymmetric

- 5) Well above the Curie temperature, ferromagnetic materials behave like _____ material.

a) a diamagnetic	b) a paramagnetic
c) a ferrimagnetic	d) an antiferromagnetic

- 6) Magnetic materials which can be readily magnetized in either direction are called _____.
 - a) soft magnetic materials
 - b) hard magnetic materials
 - c) low hysteresis loss materials
 - d) high hysteresis loss materials

Q.5 Answer the following. (Any Two)

12

- a) Explain in detail motion of electron through the solid using Kronig - Penney model.
- b) Explain Langevin's theory of paramagnetism.
- c) Obtain London equation in superconductivity.

B) Fill in the blanks OR Write True/False: 04

- 1) The gain of an open-loop op-amp is usually _____.
- 2) True/False: The output of an inverting integrator decreases when the input voltage is positive.
- 3) True/False: JK master-slave flip-flop eliminates the race-around condition.
- 4) Multiplexer is used to _____ signals.

Q.2 Answer the following. (Any Six) 12

- a) Draw the AC equivalent circuit of a differential amplifier.
- b) Explain the concept of negative feedback in op-amps.
- c) State the advantages of synchronous counters over asynchronous counters.
- d) Define input bias current and offset current.
- e) List various applications of op-amp.
- f) Write the truth table of D Flip-Flop.
- g) Explain the concept of voltage series feedback.
- h) Write an assembly language program for 8085 to subtraction of two 4-bit numbers.

Q.3 Answer the following. (Any Three) 12

- a) Explain the voltage follower and its applications.
- b) Draw and explain instrumentation amplifier using op-amps.
- c) Explain operation of Astable multivibrator using op-amp.
- d) Describe DC analysis of an differential amplifier.

Q.4 Answer the following. (Any Two) 12

- a) Explain the working of phase shift oscillator using op-amp.
- b) Explain the architecture of 8085 microprocessor.
- c) Design a differentiator circuit using op-amp.

Q.5 Answer the following. (Any Two) 12

- a) Explain the working of JK Flip-Flop with truth table.
- b) Write short notes on Wein bridge oscillator.
- c) Write an Assembly Language program for 8085 to multiply two 4-bit numbers.

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Set **P**

**M.Sc. Physics (Solid State) (Semester - I) (New) (NEP CBCS)
Examination: March/April – 2026
Elements of Materials Science (2307107)**

Day & Date: Wednesday, 22-04-2026
Time: 03:00 PM To 05:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative.**08**

- 1) The material which absorbs the transmission of visible light is termed as _____.
 - a) Opaque
 - b) Translucent
 - c) Transparent
 - d) None of the above
- 2) The forces exerted between the atoms/molecules are _____ in nature and depend on the electronic structure.
 - a) Electrostatic
 - b) Electrodynamic
 - c) Ionic
 - d) Electromagnetic
- 3) _____ is the process of increase in the electrical conductivity of a semiconducting material, when a radiation falls on the material.
 - a) Photodetectivity
 - b) Photoconductivity
 - c) Specific detectivity
 - d) All the above
- 4) The degree of freedom when ice, water and water vapour co-exist in equilibrium is _____.
 - a) 1
 - b) triple point
 - c) zero
 - d) -1
- 5) Due to the cross-linking of polymer chain, the strength of the polymer increases while its _____ decreases.
 - a) photodetectivity
 - b) conductivity
 - c) elasticity
 - d) plasticity
- 6) _____ are organic materials prepared by polymerization reactions in which small molecules are chemically combined into long chain molecules or 3D structures.
 - a) Ceramic
 - b) Metals
 - c) Alloys
 - d) Polymers
- 7) The unit of flux J is _____.
 - a) atoms $m^{-2} s^{-1}$
 - b) atoms $m^2 s^{-1}$
 - c) moles $m^{-3} s^{-1}$
 - d) All of above

- 8) The linear polymers are held together by weak _____ forces.
- a) ionic
 - b) metallic
 - c) van der Waals
 - d) covalent

B) Write True or False. 04

- 1) The amorphous solids have no regular structure (no directional property) and hence they are known as isotropic substances.
- 2) The nanoparticles with particle size of nearly 1-100 nm.
- 3) Responsivity is defined as the ratio of the electrical output to the radiation input.
- 4) The unit of the diffusion coefficient D is $\text{m}^2 \text{s}^{-1}$.

Q.2 Answer the following. (Any Six) 12

- a) Explain skin depth.
- b) What are the primary bonds?
- c) What is degree of polymerization?
- d) How can luminescence be classified?
- e) What is the degree of freedom of a system of two components when the number of phases is one?
- f) What are semiconducting materials?
- g) What is the phase rule?
- h) What are the key characterization techniques for nanostructured materials?

Q.3 Answer the following. (Any Three) 12

- a) What chemical techniques are used for the synthesis of nanophase materials? Describe any two of these in short.
- b) Explain Kirkendall effect.
- c) Discuss interaction of light with matter.
- d) Explain the difference between crystalline and non-crystalline (amorphous) solids.

Q.4 Answer the following. (Any Two) 12

- a) What is principle of luminescence? Explain classification of luminescence.
- b) What are the types of Engineered materials? Discuss in short. What are the application of engineering materials?
- c) What are nanostructured or nanophase materials, and what are their important properties and applications?

Q.5 Answer the following. (Any Two) 12

- a) What are ceramics, and how are ceramics classified into different categories?
- b) What is the photoconductivity? What are the characteristics of photoconductive materials?
- c) What are secondary bonds, and what are the different types of these bonds?

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Set **P**

M.Sc. Physics (Solid State) (Semester - I) (New) (NEP CBCS)
Examination: March/April - 2026
Research Methodology in Physics (2307105)

Day & Date: Friday, 24-04-2026
 Time: 03:00 PM To 05:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
 2) Figures to the right indicate full marks.

Q.1 A) Choose the correct alternatives. 08

- 1) In UV-visible spectroscopy, the concentration of the analyte can be determined using _____.
 - a) Ohm's law
 - b) Beer-Lambart's law
 - c) Lenz's law
 - d) Faraday's law
- 2) Identifying gap areas in research are for _____.
 - a) exploration of uncharted terrain
 - b) addressing unanswered queries
 - c) resolution of conflicting findings
 - d) all of these
- 3) Gathering knowledge for knowledge's sake is termed _____ research.
 - a) basic
 - b) applied
 - c) quantitative
 - d) experimental
- 4) A critical literature review is for _____.
 - a) analytical exploration
 - b) comparative evaluation
 - c) identification of patterns
 - d) all of these
- 5) The principle on which electrodeposition method works is _____.
 - a) electrolysis
 - b) evaporation
 - c) gel formation
 - d) none of these
- 6) _____ is used as the source of infrared light.
 - a) tungsten filament
 - b) mercury lamp
 - c) diode laser
 - d) global filament
- 7) Research in common parlance refers to a search for _____.
 - a) knowledge
 - b) goods
 - c) peace
 - d) skills
- 8) Atomic force microscopy measures _____ between the probe tip and the sample surface.
 - a) length
 - b) pressure
 - c) interaction force
 - d) temperature

- B) Write True or False.** **04**
- 1) Physicists must adhere to ethical guidelines when conducting research. (True/ False)
 - 2) Applied research aims at finding a solution for an immediate problem facing a society. (True/ False)
 - 3) Research methodology is a way to systematically solve the research problem. (True/ False)
 - 4) Radiofrequency sputtering can sputter insulating materials. (True/ False)
- Q.2 Answer the following (Any Six)** **12**
- a) Define research.
 - b) Write use of web in research.
 - c) Draw the neat labelled schematic diagram of scanning tunneling microscope.
 - d) Write the advantages and disadvantages of UV-visible spectroscopy.
 - e) What is the importance of literature review in research?
 - f) Write the names of tools used in research for data collection.
 - g) What are the types of sputtering?
 - h) Write the advantages of electrodeposition method.
- Q.3 Answer the following. (Any Three)** **12**
- a) Explain Fourier transform infrared spectroscopy with neat labelled diagram.
 - b) Explain in detail construction and working of Magnetron Sputtering.
 - c) Explain the concept data interpretation in research.
 - d) Write a note on meaning and importance of research.
- Q.4 Answer the following. (Any Two)** **12**
- a) Explain data analysis in research with statistical tools.
 - b) Explain in detail spray pyrolysis deposition technique.
 - c) Explain in details types of research.
- Q.5 Answer the following. (Any Two)** **12**
- a) Explain transmission electron microscope with neat labelled diagram.
 - b) Explain in detail ion beam sputtering technique of deposition.
 - c) Write detailed note on atomic force microscopy.

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Set **P**

**M.Sc. Physics (Solid State) (Semester - II) (New) (NEP CBCS)
Examination: March/April – 2026
Quantum Mechanics (2307201)**

Day & Date: Thursday, 16-04-2026
Time: 11:00 AM To 01:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative. (MCQ)**08**

- 1) If Ψ_a and Ψ_b are orthogonal to each other, then scalar product is _____.
 - a) one
 - b) zero
 - c) infinite
 - d) Ψ_a
- 2) The product of a scalar with a vector gives _____.
 - a) another vector
 - b) scalar
 - c) pseudo scalar
 - d) null vector
- 3) The uncertainty relation cannot hold for the following pairs _____.
 - a) position and momentum
 - b) energy and time
 - c) linear momentum and angle
 - d) angular momentum and angle
- 4) Which of the following is an even function?
 - i) $|x| - 1$
 - ii) $-x^2 + 2$
 - iii) $\cos x$
 - iv) $\frac{1}{x^2}$
 - a) i) & ii)
 - b) ii) & iii)
 - c) iii) & iv)
 - d) All of these
- 5) Energy of a particle which is moving in one-dimensional rigid box is proportional to _____.
 - a) square of length of box
 - b) reciprocal of square of length of box
 - c) length of box
 - d) reciprocal of length of box
- 6) The total energy operator or Hamiltonian operator is given by $\hat{H} =$ _____.
 - a) $\frac{\hat{p}^2}{2m}$
 - b) $V(x)$
 - c) $\frac{\hat{p}^2}{2m} + V(x)$
 - d) $\frac{\hat{p}^2}{2m} - V(x)$

Q.5 Answer the following. (Any Two)

12

- a) State and explain 'Heisenberg's uncertainty principle.'
- b) Obtain Schrodinger's wave equation for a square well potential and obtain its solution.
- c) Describe the Pauli spin matrices.

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Set **P**

M.Sc. Physics (Solid State) (Sem - II) (New) (NEP CBCS)
Examination: March/April – 2026
Electrodynamics (2307202)

Day & Date: Saturday, 18-04-2026
 Time: 11:00 AM To 01:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
 2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative. (MCQ)**08**

- 1) The Biot-Savart law gives the magnetic field due to _____.
 - a) Point charge
 - b) Dipole moment
 - c) Current element
 - d) Magnetic flux
- 2) In an electromagnetic wave, the direction of magnetic field induction B^{\rightarrow} is _____.
 - a) parallel to electric field E^{\rightarrow}
 - b) perpendicular to electric field E^{\rightarrow}
 - c) random
 - d) none of the above
- 3) Lienard-Wiechert potentials are associated with: _____.
 - a) Static charges
 - b) Moving charges
 - c) Magnetic monopoles
 - d) Neutral particles
- 4) The Poynting's vector S of an electromagnetic wave is _____.
 - a) $S = \vec{E} \times \vec{H}$
 - b) $S = \vec{E} \times \vec{B}$
 - c) $S = E/B$
 - d) $S = E/H$
- 5) In a stationary medium, EM wave propagation depends on: _____.
 - a) Mass
 - b) Charge
 - c) Permittivity and permeability
 - d) Conductivity
- 6) The total power radiated by an oscillating dipole is _____ to _____ power of frequency.
 - a) Proportional, fourth
 - b) Inversely proportional, fourth
 - c) Inversely proportional, third
 - d) Proportional, third
- 7) _____ of the following laws do not form a Maxwell equation.
 - a) Planck's law
 - b) Gauss's Law
 - c) Faraday's law
 - d) Ampere's Law

- 8) Poynting's vector S gives _____.
 a) Energy transported per unit area per second
 b) Energy stored per unit volume
 c) Flux of field
 d) Electromagnetic Momentum contained per unit volume

B) Write True /False:**04**

- 1) The electrostatic potential energy of a system depends on the configuration of charges. (True/False)
- 2) Reflection coefficient can be greater than one. (True/False)
- 3) In equipotential surface, potential is different everywhere. (True/False)
- 4) Divergence theorem is based on Gauss law. (True/False)

Q.2 Answer the following. (Any Six)**12**

- a) State and write expression for Biot-Savart law.
- b) What is the meaning of divergence of electric field?
- c) What are the scalar and vector potentials?
- d) State and write expression for Faraday's law.
- e) Define skin depth.
- f) Define magnetic dipole.
- g) Define electromagnetic wave.
- h) What is D'Alembertian operator?

Q.3 Answer the following. (Any Three)**12**

- a) Explain Maxwell's equations in differential form in free space.
- b) Write a note on energy stored in electric and magnetic fields.
- c) Write a note on Lorentz's and Coulomb's gauges.
- d) Explain radiation from a half wave antenna.

Q.4 Answer the following. (Any Two)**12**

- a) Derive an expression for differential form of Gauss's law.
- b) Derive an expression for Lienard-Wiechert potentials of a point charge.
- c) Derive the expressions for reflection and refraction of electromagnetic waves at plane boundaries for normal incidence.

Q.5 Answer the following. (Any Two)**12**

- a) State and prove Poynting's theorem.
- b) Derive an expression for differential form of Ampere's law.
- c) Discuss the interaction energy between two current loops and derive an expression for the force between them.

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Set **P**

**M.Sc. Physics (Solid State) (Semester - II) (New) (NEP CBCS)
Examination: March/April – 2026
Classical Mechanics (2307206)**

Day & Date: Tuesday, 21-04-2026
Time: 11:00 AM To 01:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative. (MCQ)**08**

- 1) According to Hamilton's principle, the action integral for monogenic, conservative system should produce _____ value.
 - a) one
 - b) Zero
 - c) minus one
 - d) Extremum
- 2) The configuration space involves _____.
 - a) 2N dimensions
 - b) 3N dimensions
 - c) 6N dimensions
 - d) 4N dimensions
- 3) If the total force is zero, then _____ is conserved.
 - a) angular momentum
 - b) force
 - c) linear momentum
 - d) torque
- 4) The gyroscopic forces are _____ in nature.
 - a) conservative
 - b) non-conservative
 - c) pseudo
 - d) not exist
- 5) In central force laws, if the potential energy, $V = -k/r$, then _____.
 - a) $f = -k/r^2$
 - b) $f = k/r^2$
 - c) $f = k/r$
 - d) $f = -k/r$
- 6) The Poisson's bracket, $[q, q] =$ _____.
 - a) p
 - b) Q
 - c) 0
 - d) Q
- 7) The square of the period of revolution of a planet around the sun is proportional to _____.
 - a) cube root of the semi major axis of the ellipse
 - b) cube root of the semi minor axis of the ellipse
 - c) cube of the semi minor axis of the ellipse
 - d) cube of the semi major axis of the ellipse
- 8) If the transformation is canonical, the Poisson's bracket has _____ when evaluated with respect to any canonical set of variables.
 - a) same value
 - b) different value
 - c) zero value
 - d) no value

- B) Write true/false** **04**
- 1) An angular momentum is conserved in the absence of external force. (True/False)
 - 2) The constraints expressed as equations of coordinates are called holonomic constraints. (True/False)
 - 3) The fundamental Poisson's brackets are varying under canonical transformation. (True/ False)
 - 4) In δ variation, both position and time coordinate at the end points, are held fixed. (True/False)
- Q.2 Answer the following. (Any Six)** **12**
- a) Prove linear momentum is conserved for a system of particles.
 - b) Write two advantages of Lagrangian formulation over Newtonian formulation.
 - c) Define scleronomic and rheonomic constraints.
 - d) What are the generalized coordinates?
 - e) State Hamilton's principle and write its expression.
 - f) Define Poisson's bracket and write its expression.
 - g) What are the gyroscopic forces?
 - h) How many forms of generating function? Write it.
- Q.3 Answer the following. (Any Three)** **12**
- a) State and prove work-energy theorem.
 - b) Explain the characteristics of the conservative central force.
 - c) Explain any four properties of Poisson's brackets.
 - d) Deduce Lagrange's equation of motion from Hamilton's principle.
- Q.4 Answer the following. (Any Two)** **12**
- a) Explain the Guage invariance in Lagrangian formulation.
 - b) State and derive the expression for the Kepler's first law.
 - c) Deduce Euler-Lagrange's differential equation using variational technique.
- Q.5 Answer the following. (Any Two)** **12**
- a) Derive Hamilton's canonical equations of motion in terms of Poisson's brackets.
 - b) Explain the principle of least action and prove it.
 - c) Derive an expression for reduction of two body problem in to equivalent one body problem with its special case of $m_1 \gg m_2$.

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M.Sc. Physics (Solid State) (Semester - III) (New) (NEP CBCS)
Examination: March/April – 2026
Statistical Physics (2307301)

Day & Date: Friday, 17-04-2026
 Time: 11:00 AM To 01:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
 2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative. (MCQ)

08

- 1) According to law of thermodynamics the energy of the universe is always _____.
 - a) conserved
 - b) increasing
 - c) decreasing
 - d) zero
- 2) Work done by ideal gas during isothermal expansion is $W =$ _____.
 - a) $S_2 - S_1$
 - b) $T(S_2 - S_1)$
 - c) $R \log (V_2/V_1)$
 - d) Q/T
- 3) Which of the following is boson?
 - a) Electron
 - b) Positron
 - c) Proton
 - d) Photon
- 4) In canonical ensemble, the system exchange _____.
 - a) only matter
 - b) only energy
 - c) both matter and energy
 - d) neither matter nor energy
- 5) In Bose Einstein Condensation all the particle accumulate in _____.
 - a) excited state
 - b) meta state
 - c) ground state
 - d) all excited state
- 6) Phase space is _____ dimensional space.
 - a) 1
 - b) 2
 - c) 3
 - d) 6
- 7) The first order phase transitions are accompanied by a discontinuous change in _____.
 - a) Gibb's molar free energy function
 - b) Internal energy
 - c) Crystal symmetry
 - d) None of the above
- 8) Sackur-tetrode formula gives the relation for _____.
 - a) entropy
 - b) enthalpy
 - c) internal energy
 - d) Gibb's energy

B) Write True/False. 04

- 1) Electrons obey Bose Einstein Statistics.
- 2) Planck's radiation law can be derived using Fermi Dirac Statistics.
- 3) Phase equilibrium curve terminates at Critical point.
- 4) For the stable state of the system G i.e. Gibbs's free energy should be large.

Q.2 Answer the following. (Any Six) 12

- a) State 1st and 2nd law of thermodynamics.
- b) Distinguish between different types of ensembles.
- c) Draw a neat phase diagram for one component system.
- d) Explain microstates and macrostates.
- e) Write conditions for thermal equilibrium.
- f) Define critical point.
- g) Distinguish between 1st order and 2nd order phase transition.
- h) Write equation of reduced state.

Q.3 Answer the following. (Any Three) 12

- a) Derive Clausius- Clapeyron equation for first order phase transition.
- b) Obtain expression for change in entropy with change in volume.
- c) Derive Ehrenfest's equation for second order phase transition.
- d) Explain phase diagram using PT diagram.

Q.4 Answer the following. (Any Two) 12

- a) State and prove Liouville's theorem.
- b) What is Gibb's paradox? How it is resolved?
- c) Using Vander Waal's equation of reduced state, calculate the values of critical constants.

Q.5 Answer the following. (Any Two) 12

- a) Explain the second order phase transition with an example of BaTiO₃.
- b) Derive the expression for energy fluctuation in canonical ensemble.
- c) Explain in details Bose Einstein Condensation.

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Set P

M.Sc. Physics (Solid State) (Semester - III) (New) (NEP CBCS)
Examination: March/April - 2026
Atomic and Molecular Physics (2307302)

Day & Date: Monday, 20-04-2026
 Time: 11:00 AM To 01:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
 2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative. (MCQ)**08**

- 1) In the context of selection rules, which of the following changes is allowed for an electric dipole transition?
 - a) $\Delta l = 0$
 - b) $\Delta s = 1$
 - c) $\Delta j = 0, \pm 1$
 - d) $\Delta ml = 2$
- 2) The probability of finding an electron in an atom is described by _____.
 - a) Schrodinger equation
 - b) Heisenberg uncertainty principle
 - c) Pauli exclusion principle
 - d) Quantum tunneling
- 3) Which effect demonstrates the particle nature of photons?
 - a) Stark effect
 - b) Zeeman effect
 - c) Compton effect
 - d) Doppler effect
- 4) The strong field Stark effect in hydrogen results in energy splitting that is _____.
 - a) Linear with electric field strength
 - b) Quadratic with electric field strength
 - c) Independent of electric field strength
 - d) Magnetic field dependent
- 5) In an anharmonic oscillator, the energy levels are _____.
 - a) Equidistant
 - b) Non equidistant, higher energy levels spaced further apart
 - c) Completely degenerate
 - d) Discrete and equally spaced
- 6) Which of the following types of interaction is responsible for the attraction between molecules in a gas?
 - a) Ionic interaction
 - b) Covalent bonding
 - c) Van der Waals forces
 - d) Electrostatic force

- 7) Nuclear magnetic resonance (NMR) is primarily used to study _____.
a) Electron spin transitions b) Nuclear spin transitions
c) Rotational energy levels d) Molecular dissociation
- 8) Born Oppenheimer approximation is useful because it allows the separation of _____.
a) Electronic and nuclear motions in molecules
b) Nuclear spin and electron spin
c) Vibrational and rotational spectra
d) Chemical and physical properties

B) Write true or false.**04**

- 1) The azimuthal quantum number l determines the spin orientation of the electron.
- 2) The Stark effect can be observed in atoms exposed to an external magnetic field.
- 3) Spherical top molecules have identical moments of inertia along all axes.
- 4) In anharmonic oscillator the vibrational energy levels are non-equidistant.

Q.2 Answer the following. (Any Six)**12**

- a) What is the magnetic quantum number, and what does it signify about an electron's orbital?
- b) Describe the relationship between electron spin and magnetic moment.
- c) Explain the Paschen Back effect for two electron systems.
- d) State the Stark effect for a hydrogen atom.
- e) What is the spectrum of a nonrigid rotator?
- f) Explain its significance in rotational spectra.
- g) Explain the origin of hyperfine structure in atomic spectra?
- h) What is the Born Oppenheimer approximation.

Q.3 Answer the following. (Any Three)**12**

- a) Discuss the principle behind the chemical shift and its role in identifying functional groups in molecules.
- b) Discuss the classification of molecules into linear, symmetric tops, spherical tops, and asymmetric tops with example.
- c) Describe the significance of electron spin in creating magnetic moments.
- d) Explain the molecular interactions of covalent, ionic, and Van der Waals interactions and how these forces influence the behavior of molecules?

Q.4 Answer the following. (Any Two) 12

- a) Describe the experimental setup to observe the Compton shift and how this effect provides evidence for the quantization of electromagnetic radiation.
- b) Explain the relativistic corrections of hydrogen atom and discuss the Dirac equation with concept of spin-orbit interaction.
- c) Describe the Frank-Condon principle and its significance in understanding the electronic transitions in molecule.

Q.5 Answer the following. (Any Two) 12

- a) Explain the concept of quantum numbers and their significance in determining the energy and wavefunction of an electron in an atom. Discuss how each quantum number (n , l , m , s) affects the atomic orbitals?
- b) What is Born - Oppenheimer approximation? Explain how it allows the separation of electronic and nuclear motion in molecule.
- c) Explain in detail the techniques and instrumentation of microwave spectroscopy.

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Set **P**

M.Sc. Physics (Solid State) (Semester - III) (New)
(NEP CBCS) Examination: March/April - 2026
Semiconductor Physics (2307309)

Day & Date: Wednesday, 22-04-2026
 Time: 11:00 AM To 01:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
 2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative. (MCQ) 08

- 1) The crystal growth technique that allows atomic level control is _____.
 - a) CVD
 - b) MBE
 - c) Floating zone
 - d) Czochralski
- 2) Compared to a p-n junction diode, a Schottky diode has _____.
 - a) Higher forward voltage drop
 - b) Slower switching speed
 - c) Lower forward voltage drop
 - d) Wider depletion region
- 3) Which of the following is NOT a dopant for silicon?
 - a) Phosphorus
 - b) Arsenic
 - c) Boron
 - d) Copper
- 4) Electron tunneling through a potential barrier is possible because _____.
 - a) Electrons have zero mass
 - b) Energy is always conserved classically
 - c) The wavefunction decays exponentially inside the barrier
 - d) Barrier height becomes zero
- 5) The key use of Zener diode is _____.
 - a) Amplification
 - b) Voltage regulation
 - c) Current amplification
 - d) Oscillation generation
- 6) In an abrupt p-n junction under reverse bias, the depletion width _____.
 - a) Increases with applied reverse voltage
 - b) Is independent of voltage
 - c) Decreases linearly with voltage
 - d) Becomes zero
- 7) Space charge in a junction is formed by _____.
 - a) Neutral atoms
 - b) Ionized dopants
 - c) Free electrons
 - d) Holes only

- 8) The periodic table is arranged according to _____.
a) Atomic number b) Atomic mass
b) Group number d) Atomic radius

B) Write True or False. 04

- 1) Schottky diodes exhibit fast switching speed.
- 2) Zener breakdown occurs mainly due to quantum mechanical tunneling.
- 3) Photolithography is used to grow crystals.
- 4) In equilibrium, the Fermi level remains constant throughout a p-n junction.

Q.2 Answer the following. (Any Six) 12

- a) Why is silicon the dominant semiconductor material?
- b) Draw cubic lattice diagrams for (111) and (001) planes.
- c) Explain rectification principle of p-n junction diode.
- d) Describe the types of junction capacitances.
- e) What is a Schottky barrier?
- f) Explain the Bohr postulates.
- g) Explain the purpose of rapid thermal processing.
- h) What is the uncertainty principle?

Q.3 Answer the following. (Any Three) 12

- a) Describe crystal lattice types with diagrams.
- b) Explain ion implantation process and its merits.
- c) Give review of quantum numbers.
- d) Discuss Zener breakdown and avalanche breakdown mechanisms in p-n junctions.

Q.4 Answer the following. (Any Two) 12

- a) Explain photolithography steps in device fabrication.
- b) Explain the diamond lattice and its importance in electronics.
- c) Explain graded junction characteristics.

Q.5 Answer the following. (Any Two) 12

- a) Explain MBE in detail and its advantages.
- b) Derive energy of electron in Bohr's nth orbit.
- c) Derive Schrodinger's equation in 1-D potential well.

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M.Sc. Physics (Solid State) (Semester - IV) (New) (NEP CBCS)
Examination: March/April - 2026
Physics of Semiconductor Devices (2307401)

Day & Date: Thursday, 16-04-2026
 Time: 03:00 PM To 05:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
 2) Figures to the right indicate full marks.

Q.1 A) Choose the correct alternative. (MCQ) 08

- 1) Germanium has a band gap of _____ eV.
 - a) 1.12
 - b) 0.66
 - c) 5.6
 - d) 10
- 2) The two classes of semiconductor energy bands are _____.
 - a) direct and indirect
 - b) direct and individual
 - c) indirect and individual
 - d) direct and imaginary
- 3) How much should be minimum energy of the photon in joule that can be absorbed by semiconductor having band gap of 1.6 eV?
 - a) 2 joule
 - b) 1.6×10^{-19} joule
 - c) 3.2×10^{-19} joule
 - d) 2.56×10^{-19} joule
- 4) For optical absorption, the energy of the impinging photon _____.
 - a) $h\nu < E_g$
 - b) $h\nu = \frac{E_g}{2}$
 - c) $h\nu \geq E_g$
 - d) $h\nu = \frac{E_g}{5}$
- 5) Normally, the junction area of a Schottky diode is small therefore, the junction capacitance is _____.
 - a) large
 - b) ideally infinite
 - c) very large
 - d) small
- 6) In thermionic emission _____.
 - a) A very strong electric field is applied to the metal
 - b) A metal is heated to a sufficiently high temperature
 - c) Light of a certain frequency is made to fall on the metal surface
 - d) electrons are emitted from a metal surface due to bombardment by high-speed electrons
- 7) After the SCR is fired, it will remain ON until its current drops below _____ current.
 - a) holding
 - b) transient
 - c) latch
 - d) maximum

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**M.Sc. Physics (Solid state) (Sem - IV) (New) (NEP CBCS)
Examination: March/April - 2026
Nuclear and Particle Physics (2307402)**

Day & Date: Saturday, 18-04-2026
Time: 03:00 PM To 05:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative. (MCQ) 08

- 1) The decay constant (λ) of a radioactive sample depends on _____.
 - a) the number of atoms in the sample
 - b) mass number of the nucleus
 - c) the half-life of the sample
 - d) atomic number

- 2) The asymmetry term in semi empirical mass formula is due to __.
 - a) Non equality of proton number Z and neutron number N
 - b) Non spherical shape of nuclei
 - c) Charge independence of nuclear force
 - d) Non zero value of quadrupole moment of nuclei

- 3) The _____ have explained the magic numbers.

a) Liquid drop model	b) Fermi gas model
c) Shell model	d) All of these

- 4) Two-neutron separation energy is designated as _____.

a) S_2	b) S_{2n}
c) 2_nS	d) S^{2n}

- 5) The particles having _____ spin quantum number are called fermions.

a) 0	b) $\frac{1}{2}$
c) 1	d) 2

- 6) When hydrogen atom loses an electron, it becomes _____.

a) proton	b) positron
c) electron	d) neutron

- 7) The nuclear process in which one or more particles may be liberated when the target nucleus absorbs γ - ray is called _____.

a) photodisintegration	b) photoelectric effect
c) Compton absorption	d) γ -decay

- 8) Cyclotrons are specially used to accelerate _____.
 a) negative ion b) positive ion
 c) neutron d) electron

B) Write True /False: 04

- 1) The electric quadrupole moment is negative; shape of the nuclei is oblate. (True/False)
- 2) The Rutherford scattering experiment may not be called a nuclear reaction. (True/False)
- 3) Parity can provide information about the nuclear structure. (True/False)
- 4) Cosmic rays are low-energy particles that originate in our solar system. (True/False)

Q.2 Answer the following. (Any Six) 12

- a) Write a note on shape and size of nucleus.
- b) What is an individual particle model?
- c) Why the neutron well is deeper than the proton well in nuclear potential?
- d) What is the meaning of Fermi gas in Fermi gas nuclear model?
- e) What is Q-value of nuclear reaction?
- f) Write an example of photo disintegration reaction.
- g) What is convenient unit of nuclear cross section? Convert it into SI system.
- h) What is the name of exchanging particle in the electromagnetic interaction?

Q.3 Answer the following. (Any Three) 12

- a) Explain binding energy and nuclear stability.
- b) Explain Bethe-Weizsacker mass formula.
- c) Explain in detail nuclear fission.
- d) Write a note on synchrotron.

Q.4 Answer the following. (Any Two) 12

- a) What is radioactivity? Explain law of radioactivity.
- b) How shell model explained the existence of all magic numbers with the help of different nuclear potentials?
- c) Give detailed account on origin of cosmic rays and its energy distribution.

Q.5 Answer the following. (Any Two) 12

- a) Write detailed note on nuclear scattering cross section.
- b) Give detailed account on nuclear forces and its properties.
- c) Explain idea of quantum chromodynamics and different types of quarks charge.

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**M.Sc. Physics (Solid State) (Semester - IV) (New)
(NEP CBCS) Examination: March/April – 2026
Communication System (2307405)**

Day & Date: Tuesday, 21-04-2026
Time: 03:00 PM To 05:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
2) Figures to the right indicate full marks.

Q.1 A) Choose the correct alternative.

08

- 1) The class of amplifier that gives maximum efficiency is: _____.
 - a) Class A
 - b) Class B
 - c) Class C
 - d) Class AB
- 2) In FM, the frequency of carrier varies with: _____.
 - a) Modulating frequency
 - b) Amplitude
 - c) Phase
 - d) Modulating signal amplitude
- 3) The advantage of Delta modulation over PCM is _____.
 - a) Simplicity
 - b) Higher bandwidth
 - c) Less quantization noise
 - d) Complex hardware
- 4) TDMA stands for _____.
 - a) Time Division Multiple Access
 - b) Time Domain Modulation Access
 - c) Total Data Modulation Access
 - d) Time Delay Multiple Access
- 5) In AM, the bandwidth is _____.
 - a) Equal to carrier frequency
 - b) Twice the modulating frequency
 - c) Sum of carrier and modulating frequency
 - d) Half the modulating frequency
- 6) VCO is mainly used in: _____.
 - a) Amplitude modulation
 - b) Frequency modulation
 - c) Pulse modulation
 - d) Delta modulation
- 7) Which is a digital modulation technique?
 - a) AM
 - b) FSK
 - c) FM
 - d) PM
- 8) A balanced modulator suppresses: _____.
 - a) Carrier
 - b) Sidebands
 - c) Both
 - d) None

B) Fill in the blanks / True or False: 04

- 1) The AM detector circuit is used to _____ the modulated signal.
- 2) FDM stands for _____.
- 3) T/F: PAM is a digital modulation technique.
- 4) T/F: PLL can be used for FM detection.

Q.2 Answer the following. (Any Six) 12

- a) Define modulation and demodulation.
- b) What is the function of balanced modulator?
- c) Draw a block diagram of an FM transmitter.
- d) What is the principle of sampling theorem?
- e) State the applications of TDM.
- f) Draw block diagram of AM Receiver.
- g) Compare simplex and duplex communication.
- h) Define NRZ and RZ formats.

Q.3 Answer the following. (Any Three) 12

- a) Explain the working of a class B audio amplifier.
- b) Compare PAM, PWM, and PPM.
- c) Explain FM radio frequency band and its allocation.
- d) Explain the generation and demodulation of PWM.

Q.4 Answer the following. (Any Two) 12

- a) Draw and explain the block diagram of a low-level AM transmitter.
- b) With neat labeled block diagram explain FM receiver.
- c) Explain TDM with a neat diagram.

Q.5 Answer the following. (Any Two) 12

- a) State and prove sampling theorem.
- b) Differentiate between ASK, FSK, and PSK.
- c) Explain the working of a transponder in satellite communication.

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Set **P**

M.Sc. Physics (Solid State) (Semester - IV) (New) (NEP CBCS)
Examination: March/April - 2026
Advanced Techniques of Materials Characterization (2307406)

Day & Date: Tuesday, 21-04-2026
 Time: 03:00 PM To 05:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
 2) Figures to the right indicate full marks.

Q.1 A) Choose the correct alternatives.**08**

- 1) Which electron type provides surface topography in SEM?
 - a) Primary
 - b) Backscattered
 - c) Secondary
 - d) Transmitted
- 2) The universal yield curve shows the SEE yield is maximum at _____.
 - a) 1 keV
 - b) 10 eV
 - c) 100 eV
 - d) 10 keV
- 3) The basic principle of FTIR spectroscopy is based on: _____.
 - a) Absorption of UV light
 - b) Absorption of Infrared radiation
 - c) Emission of X-rays
 - d) Scattering of light
- 4) Force vs distance curve in AFM shows: _____.
 - a) Resolution limits
 - b) Scanning speed
 - c) Tip-sample interaction
 - d) Lens strength
- 5) Which lens defect causes colored fringes around objects?
 - a) Spherical aberration
 - b) Astigmatism
 - c) Chromatic aberration
 - d) Coma
- 6) Which property of light limits the resolution of a light microscope?
 - a) Intensity
 - b) Phase
 - c) Wavelength
 - d) Frequency
- 7) What is the primary mode of image formation in TEM?
 - a) Scattered light
 - b) Transmission of electrons
 - c) Emitted photons
 - d) Reflected light
- 8) Which of the following is not an energy analyzer type used in XPS?
 - a) Hemispherical analyzer
 - b) Cylindrical mirror analyzer
 - c) Quadrupole mass analyzer
 - d) Magnetic analyzer

B) Write True or False. 04

- 1) Phase contrast microscopy is useful for observing live, unstained cells.
- 2) SEM provides higher resolution than TEM.
- 3) Raman spectroscopy is a non-destructive technique.
- 4) XPS provides both qualitative and quantitative elemental analysis.

Q.2 Answer the following. (Any Six) 12

- a) Define Numerical Aperture and its significance in optical microscopy.
- b) What is the limit of resolution for a light microscope?
- c) What is the role of magnetic lenses in an electron microscope?
- d) Define Bragg's condition.
- e) Write two applications of TEM.
- f) Mention the applications of Fluorescence Microscopy.
- g) What is the difference between Stokes and -Anti-Stokes lines?
- h) Write any two applications of FTIR spectroscopy.

Q.3 Answer the following. (Any Three) 12

- a) Compare advantages of electron microscopy over light microscopy.
- b) Describe the principle of electron tunneling in STM?
- c) What is the difference between Bright Field and Dark Field imaging in TEM?
- d) Explain the principle of photoelectron emission used in XPS.

Q.4 Answer the following. (Any Two) 12

- a) Explain the classical and quantum theory of the Raman Effect.
- b) Describe the complete instrumentation of XPS, including X-ray source, vacuum system, and energy analyser.
- c) Describe different types of lens defects and methods to minimize them.

Q.5 Answer the following. (Any Two) 12

- a) Describe the operating modes of AFM (contact and non-contact).
- b) Describe the implementation of Scanning Tunneling Microscopy in instruments.
- c) Discuss the interpretation of FTIR spectra for the identification of functional groups.

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M.Sc. Physics (Solid State) (Semester - IV) (New) (NEP CBCS)
Examination: March/April – 2026
Nanomaterials Characterization Techniques (2307407)

Day & Date: Tuesday, 21-04-2026
 Time: 03:00 PM To 05:30 PM

Max. Marks: 60

Instructions: 1) All questions are compulsory.
 2) Figures to the right indicate full marks.

Q.1 A) Choose correct alternative. (MCQ) 08

- 1) X-rays can be deflected by _____.
 - a) Electric field
 - b) None of the field
 - c) Magnetic field
 - d) Electromagnetic field
- 2) Non-contact mode in AFM is typically used to _____.
 - a) Avoid damaging soft samples
 - b) Enhance contact force
 - c) Increase scanning speed
 - d) Reduce electron beam deflection
- 3) In XPS, ultra-high vacuum (UHV) is necessary to: _____.
 - a) Increase the energy of X-rays
 - b) Enhance spectral resolution
 - c) Minimize sample contamination and improve accuracy
 - d) Increase electron emission
- 4) The backscattered electrons in TEM are used primarily for _____.
 - a) Surface topography analysis
 - b) High-resolution imaging
 - c) Chemical composition analysis
 - d) Increasing electron diffraction efficiency
- 5) Lithography was invented by _____ in 1798.
 - a) Manuel Neuer
 - b) John Denver
 - c) Billy Armstrong
 - d) Alois Senefelder
- 6) AFM stands for _____.
 - a) Auto focusing microscope
 - b) Antenna focusing microscope
 - c) Atomic force microscope
 - d) None of the mentioned
- 7) Select the wavelength range corresponding to UV-visible region _____.
 - a) 2.5 μm - 1mm
 - b) 200-800 nm
 - c) 25 μm -2.5 μm
 - d) 400-800 nm

- 8) FTIR is based on the absorption of ____.
- a) Visible light
 - b) Ultraviolet rays
 - c) Microwaves
 - d) Infrared radiation

B) Write True/False. 04

- 1) AFM can image both conductive and non-conductive samples. (T/F)
- 2) High-resolution imaging of nanomaterials is only possible using optical microscopes. (T/F)
- 3) Raman spectroscopy cannot provide information about molecular vibrations. (T/F)
- 4) XRD can determine the chemical composition of a sample. (T/F)

Q.2 Answer the following. (Any Six) 12

- a) Define EDAX.
- b) Mention two advantages of TEM.
- c) Explain elastic and inelastic scattering.
- d) What is the function of STM?
- e) What is the principle of AFM?
- f) What is XRD?
- g) State any two limitations of SEM.
- h) Define Surface Tension and its relevance in nanomaterials.

Q.3 Answer the following. (Any Three) 12

- a) Write a short note on the applications of AFM.
- b) Explain Angle Resolved XPS.
- c) What is Electron Energy Analyzer.
- d) Describe the working principle of STM (Scanning Tunneling Microscope).

Q.4 Answer the following. (Any Two) 12

- a) Explain SEM in detail with neat diagram.
- b) Explain the working of UV-VIS spectroscopy and its application in nanomaterials.
- c) Write a note on Optical Absorption & Non Linear Kerr Effect.

Q.5 Answer the following. (Any Two) 12

- a) Discuss the differences, advantages, and limitations of SEM and TEM.
- b) Describes the instrumentation and working of FTIR spectrometer.
- c) Discuss quantitative analysis and applications of XPS.