# Punyashlok Ahilyadevi Holkar Solapur University, Solapur



Name of the Faculty: Science & Technology

**CHOICE BASED CREDIT SYSTEM** 

Syllabus: Entrepreneurship

Name of the Course: B.Sc. II (Sem.— III & IV)

(Syllabus to be implemented from June 2023)

# Punyashlok Ahilyadevi Holkar Solapur University, Solapur

Faculty of Science & Technology

## Choice Based Credit System (CBCS) (w.e.f. 2023-24)

Revised Structure for B.Sc. Entrepreneurship-II

Subject/ Core Course	Name and Type of the Paper		Paper/ Practical	Н	Hrs/week		Total Marks	UA	CA	Credi ts	
	Type	Nar	ne	Number	L	Т	P	Per Paper			
Class:				B.Sc II	Seme	ester –	III				
Core Courses (*Students can op	t any Thraa			Paper-V	3			50	40	10	4.0
subjects among the Subjects offered a	e Four	DSC 1C	IC-1A	Paper-VI	3			50	40	10	
OR		DSC 2C		Paper-V	3			50	40	10	4.0
Students can opt any	y Two			Paper-VI	3			50	40	10	
offered at B. Sc. I ar	subjects among the Four Subjects offered at B. Sc. I and any one from the Interdisciplinary subjects.			Paper-V	3			50	40	10	4.0
				Paper-VI	3			50	40	10	
Total SemIII	Total SemIII				18			300	240	60	12
		\$ SEC-1			4			100	80	20	4
Class:				B.Sc	II S	Semest	er –IV	7	ı		
As per SemIII		DSC 1D	IC-1B	Paper-VII	3			50	40	10	4.0
				Paper-VIII	3			50	40	10	
		DSC 2D		Paper-VII	3			50	40	10	4.0
				Paper-VIII	3			50	40	10	
		DSC 3D		Paper-VII	3			50	40	10	4.0
				Paper-VIII	3			50	40	10	
		Environment	al Studies		3			50	40	10	NC
Total Sem-IV					18			300	240	60	12
Total (Theory)		1			36			600	480	120	24
-		DSC 1C &		Pr. II &III			8	200	160	40	4.0
Core		DSC 2C &					8	200	160	40	4.0
Pract	ical	DSC 3C &	$3D^{1B}$	Pr. II & III			8	200	160	40	4.0
Total (Practicals)		1	l				24	600	480	120	24
<b>Grand Total</b>					36		24	1200	960	240	48
	\$ SEC-1				4			100	80	20	4

<sup>\*</sup>Core Courses: Chemistry/Physics//Mathematics/Statistics/Botany/Zoology/ Microbiology/ Electronics/Computer Science Geology/ Geography/Psychology

Interdisciplinary Courses - Geochemistry/Biochemistry/Meteorology/Plant Protection/NCC etc.

<sup>\$</sup> Skill Enhancement Course (SEC) the students can choose MOOCs/ NPTEL/SWAYAM/Path Shala/Add-on / Skill based courses of university/college initiated courses of same credits. (Student can register one course of minimum four credits)

<sup>\$</sup> These courses are not compulsory, but after completion of these courses students get additional credits on their mark lists. Selected Course should not be from the syllabus.

<sup>\$</sup> SEC courses run by colleges should be communicated to university for information & necessary action.

# **General Guidelines for Choice Based Credit System (CBCS)**

## B.Sc. II Entrepreneurship- Details Course structure - w. e. f. 2023-24

- 1. The University follows Semester system
- 2. An academic year shall consist of two semesters
- 3. Each B.Sc. course shall consist of three years i.e. six semesters
- 4. Environmental Studies paper shall remain compulsory for B. Sc .Part- II students in IV<sup>th</sup> Sem.
- 4. B.Sc. Part-II shall consist of two semesters: Semester III and Semester IV.

In semester –III, there will be two theory papers of 50 marks for each subject. There shall be three optional science subjects. Similarly, in semester –IV there will be two theory papers of 50 marks for each subject. There shall be three optional science subjects and Environmental Studies paper compulsory for every student in semester IV.

The scheme of evaluation of performance of candidates shall be based on University assessment as well as College internal assessment as given below. For B.Sc. Part II Sem. III & IV the internal assessment will be based on Unit tests, Home assignment, viva, practicals, Project Work etc. as given below. Practical course examination of 100 marks for each subject shall be conducted at the end of IV<sup>th</sup> semester. The practical examination of 100 marks shall also consist of 80 marks for University practical assessment and 20 marks for college internal assessment.

The process of evaluation for Environmental Studies shall be based on University theory examination of 40 marks and 10 marks internal assessment. The internal assessment for environmental studies shall be based on internal test/ home assignment/tutorial/project report of 10 marks.

For University practical examination out of two examiners, one examiner will be internal and another examiner will be External. Both examiners will be appointed by the University. The internal practical assessment shall be done as per scheme given below.

#### **5. Scheme of evaluation:**

As per the norms of the grading system of evaluation, out of 50 Marks, the candidate has to appear for College internal assessment of 10 marks and external evaluation (University Assessment) of 40 marks. The respective B.O.S. may decide the nature of College internal Assessment after referring to the scheme given below or may be used as it is.

# SOLAPUR UNIVERSITY, SOLAPUR B.Sc. II-Entrepreneurship CBCS PATTERN

w. e. f. 2023-24

# Semester-III Entrepreneurship

Paper-V- Principles of Business Management & Business Organization
Total Marks: 50 (40+10) Credits -4 Contacts hours: 45

#### **Unit I: -Business Management**

9

Definition, Nature and Importance, function, Managerial Process and roles of manager School of management & F.W. Taylor, Henry Fayola, Charles Babej, Peter Ducker, Mary Parker, Follett, Elton Mayo.

#### **Unit II: -Business Planning & Decision making**

9

Meaning & definition, Planning Process, Types of Planning, Features of Planning, Meaning & Definition, Decision making Process, Types of Decision, nature of decision and strategic decision.

#### **Unit III: - Business Organization**

9

Meaning & definition, Characteristics and Importance of organization, Types of Organization-Staff and line organization, Structure of organization – Horizontal and Vertical.

#### **Unit IV: -Direction, staffing & Controlling**

9

Meaning & definition, Characteristics and Importance of Direction, Methods of Staffing, Staff training and appraisal system, Definition of control, types of control steps in control need for control

#### **Unit V: -Leadership and Motivation**

9

Meaning & definition of motivation, Importance of motivation, Theory of motivation, Herzberg two factor theory, theory X, Y&Z, financial and non-financial incentives

Leadership: - Meaning, Importance, Functions and qualities of leader, Managerial grid and leadership style

- ➤ Business Management T. Ramaswamy
- ➤ Management Stephen P. Robbins & Marry Cowler
- ➤ Modern management Practices Dr. A.K.Gavai
- ➤ Principles and Practices of Management Amrita Singh
- ➤ Business Organization and Management- B.P.Singh &T.N Chhabr

# Entrepreneurship Paper-VI- Advanced Accountancy & Auditing

Total Marks: 50 (40+10) Credits -4 Contacts hours: 45

### **Unit I: Financial Accounting with Tally:**

10

Company creation, Ledger creation, Accounts configuration, Accounts classification, Accounts Master Creation, Voucher Types and Classes, Accounts vouchers

VAT (Value Added Tax) Introduction, Overview of GST

#### **Unit II: Final Account & Insurance Claims**

08

Bank, Bank reconciliation statement, Loss of Stock and Loss of Profit by Fire.

#### **Unit III: Fund flow & Cash flow Statement**

08

Statement of Changes in Financial, Position on Cash Basis and on Working Capital Basis.

Unit IV: Auditing 09

Meaning, nature, scope and objectives, Types of Audit – Internal Audit and External Audit, Internal Check Appointment, Qualifications, Disqualifications, Removal and Remuneration of an Auditor of a Limited Company; Statutory Audit Report.

#### Unit V: Vouching 10

Meaning, need and importance, vouching of cash and Credit transactions, Verification and Valuation of Assets and Liabilities, Special features in respect of Audit of Co-operative Societies, Bank, and Charitable Trust and Institutions.

- 1. Advanced Accountancy Shukla and Grewal.
- 2. Steps in Advanced Accountancy Maheshwari.
- 3. Principles of Management Accounting Manmohan Goyal.
- 4. Management Accounting Haneef Mukharji.
- 5. Financial Accounting Haneef Mukharji.
- 6. Tally. ERP 9(Training Guide) Ashok K. Nandani

- 7. Tally 9 Vishnu Priya Shing
- $8.\ Practical\ Auditing-B.N.\ Tandon.$
- 9. Principles of Auditing De Paula.
- 10. Principles and practice Saxena

# **Industrial Chemistry**

**Total Marks: 50 (40+10)** 

# Paper-V: Analytical and Industrial Aspects of Inorganic Chemistry

**Credits-4** 

**Contacts hours: 45** 

Unit 1: Theory of Volumetric Analysis:	09
1. Introduction, Terminology:- Titrant, Titrand, Standard solution, Titration,	Indicator, Equivalence
point, End point, Primary standard substance, Secondary standard	substance (definition,
characteristics and examples).	
2. Acid Base Titration:	
A. Introduction	
B. Colour change Interval	
C. Theory of Acid-Base indicator:	
a) Ostwald's ionization theory b) Quinoid theory	
3. Neutralization curve and choice of indicator for following titrations:	
A. Strong acid and Strong Base	
B. Strong Acid and Weak Base	
C. Weak Acid and Strong Base	
4. Complexometric titration:	
A. General account	
B. Types of EDTA Titrations (in detail direct titration)	
C. Metallo-chromic Indicator with respect to Eriochrome Black-T	
Unit II: Theory of Gravimetric Analysis:	09
1. Introduction	
2. Precipitation – Conditions of Precipitation, Physical nature of Precipitate	
3. Process of precipitation – i) Nucleation ii) Crystal growth iii) Digestion	
4. Co-precipitation and Post-precipitation	
5. Role of Organic precipitants in gravimetric analysis—	
i) DMG ii) Aluminon iii) 8- hydroxyl quinoline	
Unit III: Catalysis:	09
1. Introduction	
2. Classification of catalytic reactions: Homogeneous and Heterogeneous	
3. Types of catalyst and catalysis	
4. Characteristics of catalytic reactions	
5. Mechanism of catalysis: i) Intermediate/ Activated compound formation theorem	ory
12	

6. Industrial applications of catalysis

#### **Unit IV: Manufacture of Industrial Heavy Chemicals:**

09

- 1. Introduction, definition of heavy chemicals
- 2. Physico-chemical principle and manufacture of following chemicals:
  - a. Ammonia by Habers process.
  - b. Sulphuric acid by Contact process.
  - c. Sodium carbonate by Solvay process.

## Unit V: Corrosion and Passivity.

09

- 1. Corrosion:
  - a. Introduction, types of corrosion
  - **b.** Electrochemical theory of corrosion
  - c. Factors affecting the corrosion: i) Position of metal in emf series ii) Purity of metal
    - iii) Effect of moisture iv) Effect of oxygen v) Hydrogen over voltage
- **d.** Methods of protection of metals from corrosion
- 2. Passivity: a. Definition. b. Types of passivity. c. Oxide film theory. d. Application of passivity.

- 1. Advanced Inorganic Chemistry by Satyaprakash, Tuli, Basu (S. Chand and Co.)
- 2. Inorganic Chemistry by Puri and Sharma (S. Chand &Co.)
- 3. University General Chemistry by CNR Rao(McMillan)
- 4. Industrial Chemistry by B.K.Sharma.
- 5. Environmental Chemistry by S.M. Khopkar (Wiley EasternLtd.)
- 6. Inorganic Chemistry by D.E. Shriver, P.W. Atkins and C.H. Longford, Oxford.
- 7. Environmental chemistry by B.K.Sharma.
- 8. Text book of Quantitative Inorganic Analysis by A.I.Vogel.
- 9. Vogel's Text Book of Quantative Inorganic Analysis Bassett, Denny, Jefferyy Mendham.
- 10. Basic concepts of Analytical Chemistry by S.M. Khopkar.

# **Industrial Chemistry**

**Total Marks: 50(40+10)** 

# Paper-VI- Analytical and Industrial Aspects of Organic Chemistry

**Contacts hours: 45** 

Credits- 4

Unit I: Soaps and Detergents:	
1. Soaps:	09
i) Raw materials	
ii) Types of soaps	
iii) Manufacture of soap – Hot process	
iv) Cleansing action of soaps	
2. Detergents:	
i) Raw materials	
ii) Types of detergents - cationic, anionic, amphoteric, neutral detergents	
iii) Preparation of tee pol and deriphat	
3. Comparison between soaps and detergents.	
Unit II: Sugar and Alcohol Industry:	10
1. Manufacture of raw cane sugar	
2. Refining of raw sugar	
3. White sugar.	
4. By-products of sugar industry.	
A. Manufacture of ethyl alcohol from molasses	
B. Rectified spirit, denatured spirit, absolute alcohol and power alcohol.	
C. By-products of alcohol industry.	
Unit III: Textile Chemistry:	10
1 Introduction, classification of fibers.	
2 Sizing: i) Object of sizing, sizing ingredients and their functions.	
ii) General idea of properties of starch, softeners, synthetic adhesives.	
3 Bleaching: i) Brief study of the outline of the process of bleaching cotton and synth	etic
material.	
ii) General idea of processes like singeing, desizing, scouring.	
4 Dyeing: Study of dyeing of cellulosic material and synthetic fibers with dye like dir	ect, vat,
reactive and disperse dyes.	
Unit IV: Drugs: Synthesis and Applications:	09

- i) Antimalerials -Paludrin.
- ii) Antituberculars Isoniazide and Ethambutol.
- iii) C. N. S. drugs -Phenobarbitone.
- iv) Antidiabetic -Tolbutamide.
- v) Anti-inflammatory drugs -Ibuprofen.
- vi) Antibiotic -Chloromycetin.

#### **Unit V: Agrochemicals:**

**07** 

- 1. General idea of agrochemicals including pyrethroides
- 2. Synthesis and uses of the following agrochemicals:
  - i) Indole-3-acetic acid
  - ii) Monocrotophos
  - iii) Methoxychlor
  - iv) Ethophan
  - v) Carbaryl

#### **Reference Books:**

- Organic Chemistry R. T. Morrison and R. N. Boyd Prentice Hall of India Private limited New Delhi. 6thEdition.
- 2. A text book of Organic Chemistry ArunBahl and B. S. Bahl S. Chandand Company Ltd. 6thEdition.
- 3. Chemicals for crop improvement and pest management Green, Hartly and West.
- 4. Chemistry of pesticides K. H. Buchel (T.W.).
- 5. Medical Chemistry -Burger.
- 6. Basic Concepts of Analytical Chemistry S. M. Khopkar, Wiley EasternLtd.Bombay.
- 7. Industrial Chemistry R. K. Das, Asia Publishing, Mumbai.
- 8. Quantitative Organic Chemistry A. I. Vogel, Pearson Edn.Delhi.
- 9. Medical Chemistry A. Burger, John Viley, New York.
- 10. Biotechnology and Applied Microbiology Alani and Moo-Young.
- 11. Green Chemistry: Environment Friendly alternatives Rashmi Sanghi and M.M. Srivastava (Eds) (c) 2003 Narosa Publishing House, New Delhi, India.

Textile science - J. T.Marsh

# **Microbial Biotechnology Paper-V- Genetics**

Credits -4

Contacts hours: 45

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Unit I			08
Mendelian genetic- Introduction, Mend	lel's experiment, Monol	hybrid and Dihybrid crosses,	
Genotypic and phenotypic ratio, Law of	Dominance, Law of Inc	2	
Law of Co-dominance and Incomplete	uommance.		
Unit II			10
<b>Chromosome -</b> Structure of Chromosom	ne and Types of chromo	osome.	
<b>Chromosomal aberration</b> Translocatio	ons, inversions, deletions	s and duplications.	
Mutation- Definition, Mutagenic agent,	, Induced and Spontaneo	ous mutation.	
Unit III			06
Gene transfer			
Recombination, Transformation, Conju	gation, Transduction, F	ate of exogenote,	
Unit IV			09
Linkage: Introduction, types, phases lin	kage group detection of	f linkage significance	
Crossing over: features, theories types, f	Cactors affecting crossin	g over.	
Unit V			12
Transposable elements: -definition, typ	oes.		
<b>Gene Interaction and epitasis:</b> -Types o			
Biostatistics :Introduction, Mean, Mode	•	Null hypothesis, Chi-Square	
test, T test X2 test, Probability level and	•		

#### **REFERENCES:**

- 1. Bergey's Manual of Determinative Bacteriology- Breed and Buchanan
- 2. General microbiology Stanier

Total Marks: 50 (40±10)

- 3. General microbiology Pawar and Daginawala Vol I and II
- 4. Introduction of Biostatics.
- 5. Molecular Biology of Gene J.D. Watson
- 6. Recombinant DNA J.D. Watson
- 7. Microbiology Davis

# **Microbial Biotechnology**

# **Paper-VI- Fermentation Technology**

<b>Total Marks: 50 (40+10)</b>	Credits -4	Contacts hours: 45
Unit I		08
<b>Industrial Production of Antibiotic</b>	es:-	
Classification and Types of Antibioti	cs, Industrial Production of I	Penicillin & Streptomycin
Unit II		
<b>Industrial production of Alcoholic</b>	Beverages: -	09
Alcohol production from molasses, E	Beer Production from Barley	Malt, Wine Production from grapes.
Unit III:		10
<b>Industrial production of Enzymes-</b>	Amylase, <b>Organic acid</b> - Cit	ric Acid <b>Amino acid-</b> L-Lysine, <b>Vitamin-</b> Vit.
B <sub>12</sub> (Micro-organisms involved, prod	uction media, fermentation of	conditions, product recovery and applications)
Unit IV		10
<b>Production -</b> Single Cell Proteins		
<b>Production of Bioinsecticides</b> : - Bac	illus thuriengiensis.	
<b>Biofertilizer production</b> : - Azotobac	cter and Rhizobium	
Unit V		08
Industrial Useful Product:-Biogas	production, Biofuel production	on
Biopolymer Production- PHB and H	РНА	

# SOLAPUR UNIVERSITY, SOLAPUR B.Sc. II-Entrepreneurship CBCS PATTERN

W.e.f. 2023-24

#### **Semester-IV**

### **Entrepreneurship**

## **Paper-VII- Corporate Accounting & Professional Ethics**

Total Marks: 50 (40+10) Credits -4 Contacts hours: 45

#### **Unit I: Issue and forfeiture of Shares**

09

Issue and forfeiture of Shares, Reissue of Forfeited Shares, Valuation of Shares Valuation of Shares – Intrinsic Value Method, Market & Fair Value Method.

#### **Unit II: Final Accounts of Companies**

09

Preparation of Final Accounts of Companies in vertical form as per the Provisions of Schedule VI to the Indian Companies Act, 1956.

#### **Unit III: Liquidation of Companies**

06

Accounting for liquidation of Companies – Preparation of Liquidator's Final Statement of Account.

#### **Unit IV: Concept and Theories of Ethics & Corporate Governance**

**12** 

Meaning & Definition, Personal & Business Ethics, Morality, Etiquette & Professional codes Meaning & Definition of Corporate Governance, Corporate culture, corporate social responsibility, creating ethical organization, code of conduct.

#### **Unit V: Globalization & Functional Areas of Ethics**

**09** 

Global Corporation, Factors Facilitating Globalization, Role of MNC, Marketing Ethics, Ethics in - HRM, Financial management, IT etc.

- ❖ Business Ethics A.C.Fernando Pearson
- ❖ Business Ethics Dr. A.K.Gavai, Himalaya.
- ❖ Advanced Accountancy by M.C. Shukla, T.S. Grewal & S.C. Gupta
- Corporate Accounting by S. N. Maheshwari
- ❖ Advanced Accounting by H. Chakra borty
- Advance Accounting by Jain Narang

## Entrepreneurship

# **Paper-VIII- International Marketing & Marketing Decision**

Total Marks: 50 (40+10) Credits -4 Contacts hours: 45

#### **Unit I: - Designing Product**

08

Product Planning & Development, Product Life cycle, Product idea & its process, Product Positioning – Element of positioning, Segmentation& Targeting,

Types of Product- commodity product, technology product, customized product, Product line & product mix, Brand Management.

## **Unit II: -Pricing & Distribution**

**12** 

Definition, price decision and its objectives, Factors influence price decision, Methods of pricing, information needed for pricing, price sensitivity & price war Types of Distribution Channel, Channel Strategy,

Whole sellers – Types & Function, Retailer – Meaning & Forms- supermarket and hyper market, Physical Distribution, Marketing communication, Sales promotion, sponsorship & Exhibitions, Defining advertising strategy in competitive market, public relation and publicity

#### **Unit III: -Consumer Behavior**

08

Meaning of consumer behavior, Determinants of consumer behavior, Need of buyer, Models of behavior, buying process & Customer loyalty.

#### Unit IV: -Introduction to Global Marketing & Global marketing Environment 09

Market Selection, Emergence of Global Marketing, Use of website in marketing, Global Brand and Multinational Company, Economic, Social, political and Government, competition environment, Technology Environment.

#### **Unit V: - New Trends in Marketing**

08

Foreign Trade – steps involved in import& Export Internet marketing, E- commerce, E-marketing.

- 1. Marketing Management- V S Ramaswamy & S Namakumari
- 2. Marketing Management- Arun Kumar & N Minakshi
- 3. Global Marketing S.A.Sherlekar & V.S.Sherlekar
- 4.International marketing- Fransis Cherunitarn

# **Industrial Chemistry**

## Paper-VII: Analytical and Industrial Aspects of Physical Chemistry

Total Marks: 50(40+10) Credits -4 Contacts hours: 45

#### **Unit I: Electrochemistry:**

12

- 1. Introduction, conduction of electricity, Types of conductors: electronic and electrolytic.
- 2. Explanation of terms (Definitions and units): Conductance, Specific resistance, Specific conductance, Equivalent conductance and Molecular conductance.
- 3. Variation of specific and equivalent conductance with concentration, Equivalent conductance at infinite dilution (Mention Onsager equation,  $\lambda_V = \lambda \infty b\sqrt{\text{cgraph}}$ )
- 4. Migration of ions, Hittorf's rule, Transport number, Determination of transport number by moving boundary method, factors influencing transport number: Nature of electrolyte, concentration, temperature, complex formation and Degree of hydration.
- 5. Definition of  $p^H$  and  $p^{OH}$ , buffer solution, types of buffer, pH of buffers: Henderson's equation for acidic and basic buffers. (Derivation is not expected.)
- 6. Numerical problems.

## **Unit II: Potentiometry:**

09

- 1. Introduction.
- 2. Detail study of reference electrodes: calomel, quinhydrone and glass electrodes and their use in determination of pH.
- 3. Potentiometric titrations: Classical and analytical methods for locating end points, Advantages of potentiometric titrations,
  - i) Acid Base titrations
- ii) Redox titrations iii) Precipitation titrations.
- 4. Basic circuit of direct reading potentiometer.

#### **Unit III: Conductometry:**

**08** 

- 1. Measurement of conductance by Wheatstone bridge, Basic circuit of D.C. Wheatstone Bridge, use of alternating current, conductivity water, Different types of conductivity cells, cell constant and its determination.
- 2. Experimental determination of specific, equivalent and molecular conductances.
- 3. Conductometric acid-base titrations:
- i. Strong acid against strong base
- ii. Strong acid against weak base
- iii. Weak acid against strong base.
- Iv. Weak acid against weak base.

Unit IV Colorimetry: 08

- 1. Introduction.
- 2. General discussion of theory of colorimetry: Lambert law, Beer's law (Derivation not expected), Terms used in Colorimetry, Application of Beer's law, Deviation from Beer'slaw.
- 3. Classification of methods of 'colour' measurement or comparison, Photoelectric photometer method- single cell photo-electric colorimeter.

#### **Unit V: Flame Photometry:**

**08** 

- 1. General principles.
- 2. Instrumentation: Block diagram, Burners: Total consumption burner, premix or laminar-flow burner, Lunder garph burner, Mirrors, Slits, Monochromators, Filters and Detectors.
- 3. Applications in qualitative and quantitative analysis.
- 4. Limitations of flame photometry

- 1. Text book of Quantitative Inorganic Analysis By A. I. Vogel (ELBS and Longman 3rdEdition).
- 2. Instrumental methods of Chemical analysis by Willard, Merit and Dean.
- 3. Instrumental methods of Chemical analysis by Chatwal and Anand (Himalaya Publication).
- 4. Principles of electroplating and eletroforming by Blum and Hogaboom,
  - Mac Graw Hill Book Co. 3rdEdn.
- Vogel's text book of Quantitative Inorganic Analysis by Basssett and Denny etc. ELBS and Longman 4thEdition.
- 6. Principles of Physical Chemistry by Puri, Sharma, Pathania, Shobhanlal Naginchand and Company, Jalandar.
- 7. Text Book of Physical Chemistry by S. Glasstone, Macmillan IndiaLtd.
- 8. Elements of Physical Chemistry by D. Lewis and S. Glasstone (Macmillan).
- 9. An Introduction to Electrochemistry by S.Glasstone.
- 10. Physical Chemistry by W. J.Moore.

# **Industrial Chemistry**

# **Paper-VIII- Industrial Aspects of Applied Chemistry**

Total Marks: 50(40	+10) C	Credits- 4	Contacts hours: 45
Unit I: Metallurgy: Iron and	Steel		12
1. Introduction: - Terms used in	Metallurgy: Metallurg	gy, Mineral, Ore, G	angue, Flux, Slag
2. Occurrence of metals: Types	of Ores.		
3. Steps Involved in Metallurgio	cal Processes:		
A) Concentration of Ores:			
i) Physical Methods:-			
a) Gravity separation method	, b) Magnetic separatio	on method, c) Froth	floatation method.
ii) Chemical Methods:			
a) Calcinations b) Roasting			
B) Reduction: various methods	of reduction. Extractio	on of Iron by blast f	urnace
4. Types of steel and its alloys.			
i) Manufacture of Steel – a) l	Bessemer process b) L.	D. Process	
ii) Heat treatment on steel.			
<b>Unit II: Electroplating:</b>			09
1. Introduction.			
2. Electrolysis, Faraday's laws,	concept of Cathode an	nd Anode current e	fficiency.
3. Basic principles of electropl	ating, cleaning of articl	les.	
4. Electroplating of Nickel and	Chromium.		
5. Aluminium Anodizing proce	ess.		
<b>Unit III: Fertilizers:</b>			09
1. Classification of fertilizers.			
2. Qualities of an ideal fertilize	r <b>.</b>		
3. Manufacture of Common fer	tilizers such as:		
a. Ammonium sulphate	b. Urea		
c. Super phosphate	d. Triple superphosph	nate	
e. Potassium fertilizers			
4. Pollution caused by fertilizer	S.		

#### **Unit IV: Glass Materials:**

**08** 

- 1. Raw materials
- 2. Manufacturing methods:

Pot furnace

Tank furnace

3. Types of Glass: a. commercial Glass, b. Special glass and c. Colored glass.

#### **Unit V: Ceramic Materials:**

07

- 1. Introduction
- 2. Classification
- 3. Properties of ceramics
- 4. Cement: Types of cements and their applications
- 5. Manufacture of Portland cement by wet process.

- 1. Principles of electroplating and electroforming by Blum and Hogaboom, Mac Graw -Hill Book Co. 3rdEdn.
- 2. Vogel's text book of Quantitative Inorganic Analysis by Basssett and Dennyetc. ELBS and Longman 4th Edition.
- 3. Elements of Physical Chemistry by D. Lewis and S. Glasstone(Macmillan).
- 4. Principles of Physical Chemistry by Maron and Lando(Amerind).
- 5. An Introduction to Electrochemistry by S. Glasstone. Advanced Inorganic Chemistryby Satyaprakash, Tuli, Basu (S. Chand and Co.).
- 6. Inorganic Chemistry by G.S. Manku Tata Mc. GrawHill.
- 7. University General Chemistry by CNR Rao(McMillan).
- 8. Industrial Chemistry by B.K.Sharma.
- 9. Environmental Chemistry by S.M. Khopkar (Wiley EasternLtd.)
- 10. Industrial Chemistry: R K. Das.

# **Microbial Biotechnology**

# Paper-VII- Molecular Biology

Total Marks: 50 (40+10) Credits -4 Contacts hours: 45

Unit I 09

#### **Central Dogma**

DNA structure; Salient features of double helix; Types of DNA, the Central Dogma, Genetic code – evidences and properties.

Unit II 09

**DNA replication**- Definition, Enzyme involved in Replication, DNA Polymerases Replication in Prokaryotic Cell & Eukaryotic Cell, Rolling Circle Model

Unit III 10

**Transcription-** In Prokaryotic Cell & Eukaryotic Cell, RNA Polymerases, Post transcriptional modification

**Translation** 

Unit IV 09

Gene Regulation in Prokaryotes:

Principles of transcriptional regulation; Operon concept; Repression and Induction of genes; Regulation of Operon: Lac Operon and Trp Operon.

Unit-V 08

DNA Repair Mechanism-

Direct repair, Excision repair, Mismatch repair, SOS repair

#### **Reference Books:**

Advances in Biotechnology – S.W. Jogdand.

Textbook of Biotechnology – R.C. Dubey.

Biotechnology – B.D. Singh.

Gene VII; Benjamin Lewin; Pearson Education.

Molecular Biology; R. Weaver; 2nd Edition, McGraw Hill.

# **Microbial Biotechnology**

## Paper-VIII- Food & Dairy Technology

Total Marks: 50 (40+10) Credits -4 Contacts hours: 45

#### **Unit I: Food & Dairy Microbiology**

10

Microbiology of Food and milk, Microbial Examination of milk & food. Dye reduction tests- MBRT, Resazurin Test, Pasteurization of milk -Methods of Pasteurization – LTH, HTST, and UHT. Phosphatase test for determination of efficiency of Pasteurization. Chemical and Physical properties of food affecting microbial growth (intrinsic and extrinsic factors).

#### **Unit II Dairy Technology**

**10** 

Introduction- Dairy technology, Definition of milk, factors involved in milk contamination, Manufacturing, packaging and storage of pasteurized milk, Homogenized milk, flavored milk, Tanned milk.

Unit III 7

**Cream:** Definition, composition food & nutritive value, production and uses.

Butter: Introduction, definition, classification, composition, defect of butter uses.

Unit IV 8

**Cheese:** Introduction, definition, history, composition and types, manufacturer of cheese & its uses.

**Ice Cream:** Introduction, definition, composition, method of manufacture, packing, hardening, storage, uses.

#### **Unit V Food Technology**

**10** 

Food as substrate for microorganism, Microbial Spoilage of meat and meat product, fish & poultry foods, fruits & vegetable, General principles and different method of Preservation of food, Canned food, process of canning of food, microbial food poisoning.

#### **REFERENCE BOOKS:**

- 1) Food Microbiology (1995)-Adams M.R.and Moss, M.O., New Age International Limited.
- 2) Food Microbiology –Frazier, W.C., Westhoff, D.C. IVth edition, Tata McGraw Hill Publisher.
- 3) Industrial Microbiology by A. H. Patel, Mac Millan India Pvt. Ltd.
- 4) Modern Food Microbiology VIth edition- James M Jay. An Aspen publication.
- 5) Applied Dairy Microbiology –Elmer Marth and James Steele 2nd edition, publisher Marcel Dekker Inc.
- 6. Dairy Technology Sukumar De
- 7. Industrial Microbiology Prescott and Dunn
- 9. Industrial Microbiology Casida

#### SYLLABUS FOR LAB COURSES

# Entrepreneurship Practical based on DSC 1C & 1D Pr. II & III

## (8 periods, per week / batch)

**Total marks 100(80+20)** 

- 1. Practical Related to UNIT- I & II in Marketing
- 2. Collection of Newspaper / Magazine cuttings related to Management.
- 3. Prepare a Study Plan for academic year.
- 4. A study of Organization Structure of any Organization.
- 5. Preparation of an organization chart.
- 6. Study of selection process of any organization.
- 7. A study of buying behavior for any organization.
- 8. Design marketing research plan.
- 9. A study of product life cycle of any product.
- 10. A study of marketing channel of any company.
- 11. A visit to export unit and prepare a report.
- 12. Share Market study
- 13. Preparation of Final account of Company & Bank
- 14. Exercise on holding company
- 15. Draw a gateway of tally menu
- 16. Generalized Entries and Display Balance sheet, Cash and Bank Ledger
- 17. Problem based on Tally practical

#### Practical (visit report)

- 15) Plant and machineries in organization.
- 16) Working efficiency of organization.
- 17) Quality control & management.

# INDUSTRIAL CHEMISTRY

#### Sem. III & IV

#### Practical Based on DSC 2C & 2D

#### **Practicals- II and III**

## (08 periods, per week/batch)

**Total marks 100 (80+20)** 

#### **A Volumetric Experiments:**

- 1. Preparation of 0.1N standard solution of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> and standardize the given ferrous ammonium sulphate (FAS) solution using prepared potassium dichromate solution.
- 2. Determine the percentage of Nitrogen in the given sample of nitrogenous fertilizer (Urea or Ammonium Sulphate).
- 3. To investigate the adsorption of oxalic acid or Acetic acid from aqueous solution by activated charcoal and examine the validity of Freundlich and Longmuir isotherms.
- 4. Estimation of copper from brass by using standard sodium thiosulphate solution.
- 5. Estimation of zinc in brass solution.
- 6. Estimation of aspirin (acetyl salicylic acid).
- 7. Estimation of ethyl benzoate.
- 8. Estimation of sucrose.
- 9. Determine the chemical oxygen demand (COD) of given water sample.
- 10. Determine the biological oxygen demand (BOD) of the given water sample.
- 11. Analysis of commercial vinegar.

#### **B** Preparations:

- 1. Preparation of benzoic acid from benzamide.
- 2. Preparation of methyl orange.
- 3. Laboratory preparation of soap.
- 4. Preparation of Ferrous ammonium sulphate from ferrous sulphate.
- 5. Preparation of p-Bromo acetanilide from given acetanilide.
- 6. Preparation of tetraaminecopper(II) sulphate from copper sulphate.
- 7. Preparation of phthalimide from phthalicanhydride.

#### C GravimetricAnalysis:

1. Determine the amount of Fe as a Fe<sub>2</sub>O<sub>3</sub> from the given solution of Ferrous ammonium sulphate and

- sulphuric acid, gravimetrically.
- 2. Determine the amount of Ba as a BaSO<sub>4</sub> from the given solution of barium chloride and free hydrochloric acid, gravimetrically.
- 3. Estimation of rate of corrosion of aluminum in acidic and basic medium.

#### D Instrumental Analysis:

- 1. To Verify the Ostwald's dilution law for weak acid conductometrically.
- 2. Determine the normality of the given strong acid by titrating it against strong alkali, conductometrically
- 3. Determination of pH of the buffer solutions, potentiometrically.
- 4. Determination of dissociation constant of weak acid, pH-metrically.
- 5. To Verify the Beers-Lamberts law for copper solution and determine the concentration of given copper sample.

## MICROBIAL TECHNOLOGY Practical Based on DSC 3C & 3D

#### Pr. II & III

## (8 periods, per week / batch)

**Total marks 100(80+20)** 

- 1) Karyotypes analysis
- 2) Isolation of plasmid DNA
- 3) Isolation of Genomic DNA
- 4) Isolation of Plant DNA
- 5) Analysis for chi square test.
- 6) Study of bacterial conjugation.
- 7) Calculate mean, mode and median of the any sample.
- 8) Separation of plasmid DNA
- 9) Problem based on Mendelian genetics
  - -Law of dominance
  - -Law of Segregation
  - -Law of Independent Assortment
- 10) Isolation of Mutants
- 11) Isolation of Bacteriophage
- 12) Determination of fat from the given sample of milk.
- 13) Determination of the casein proteins from the milk sample.
- 14) Determination of sugar from the milk sample.
- 15) Determination of benzoate or sorbet content of food.
- 16) MBRT Test. & Phosphates test
- 17) MPN of milk.
- 18) Isolation and identification of Salmonella group of microorganism from milk/food.
- 19) Production of Amylase enzyme by using Bacillus species
- 20) Production of beer
- 21) Isolation antibiotic producing microorganism from soil sample.
- 22) Production of Biofertilizer- Azotobacter and Rhizobium